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The Geochemistry of Nuclear Fallout

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8 March 2018

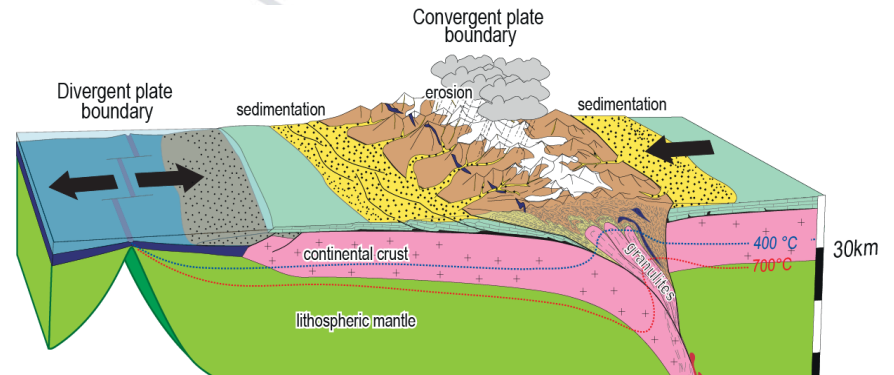
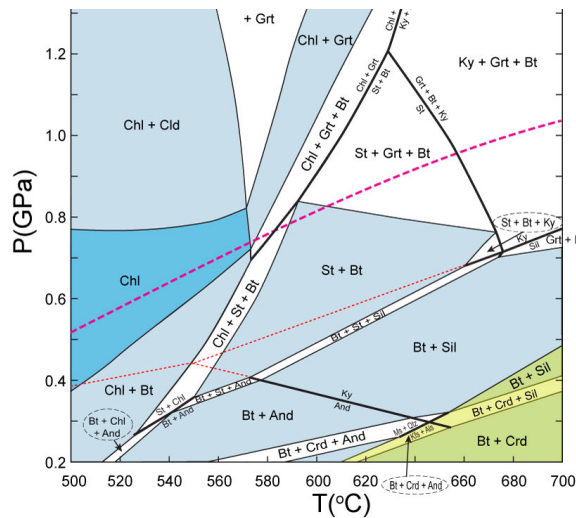
Arizona State Univ

Richard Hervig

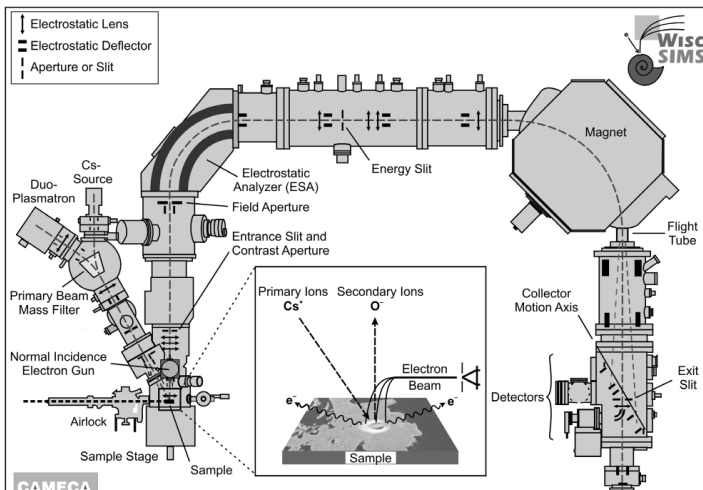
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A Geochemist's Skill Set

1. Thermodynamics & Kinetics



2. Earth Materials Processes



4. Field Work

3. Analytical Methods & Instrumentation

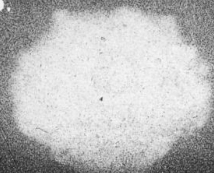
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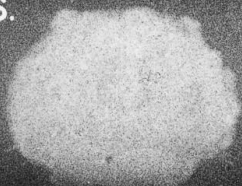
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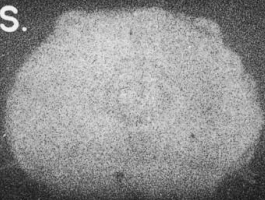
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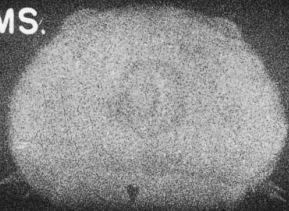
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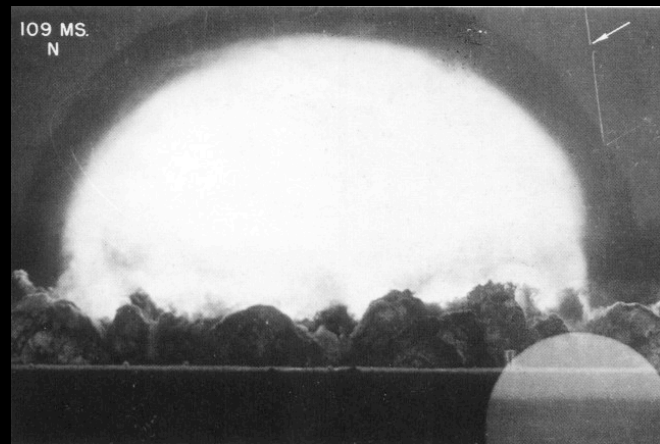
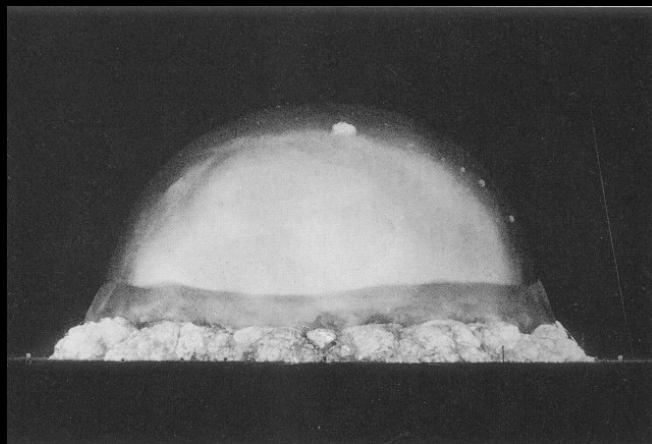
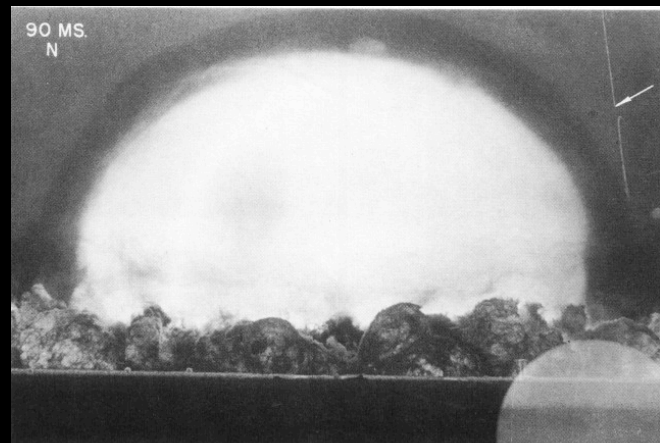
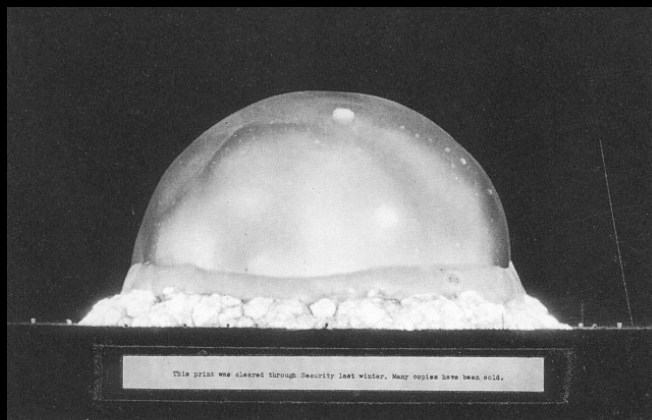
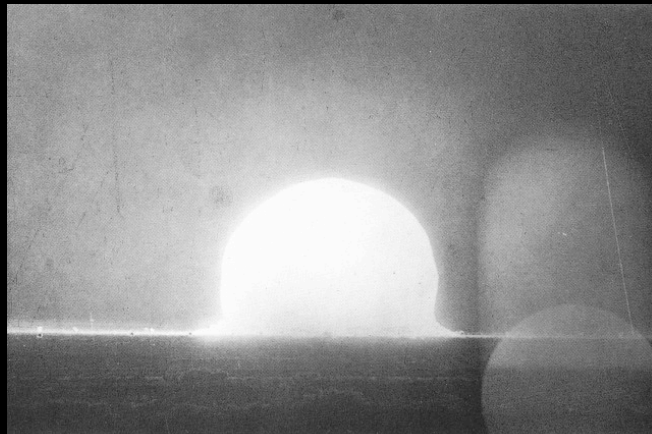
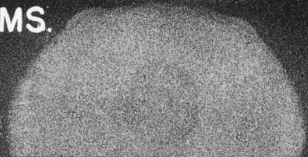
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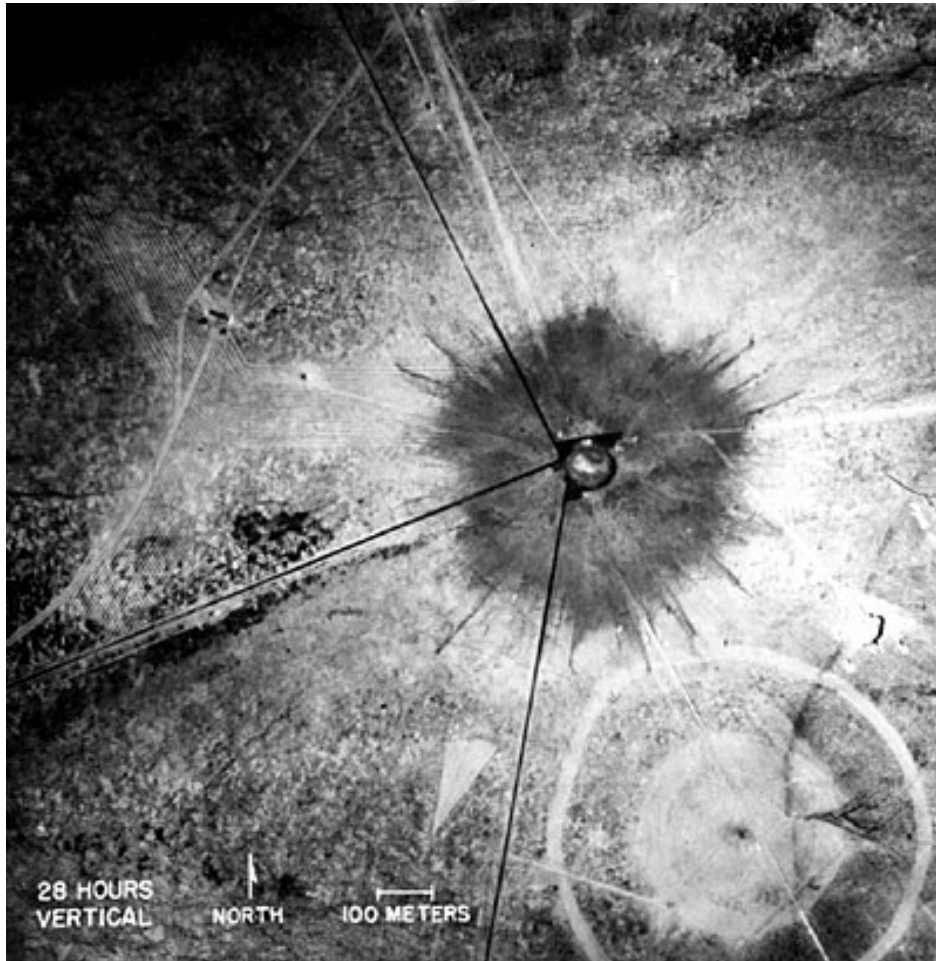
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0.94 MS.



Trinitite



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LABORATORY
943



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Trinitite: What is it Good For?

- Accessible example of nuclear explosion products
- (Cryptic) chemical and isotopic representative of an atomic bomb
- Ability to attribute a particular device to a particular source is a nuclear deterrent

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Trinity Test

- ^{239}Pu fuel packed in natural U
- Detonated from a steel tower
- Arkosic sand ground material
 - Major minerals: quartz, microcline, anorthoclase, calcite
 - Minor minerals: gypsum, clay, hornblende, salts, zircon



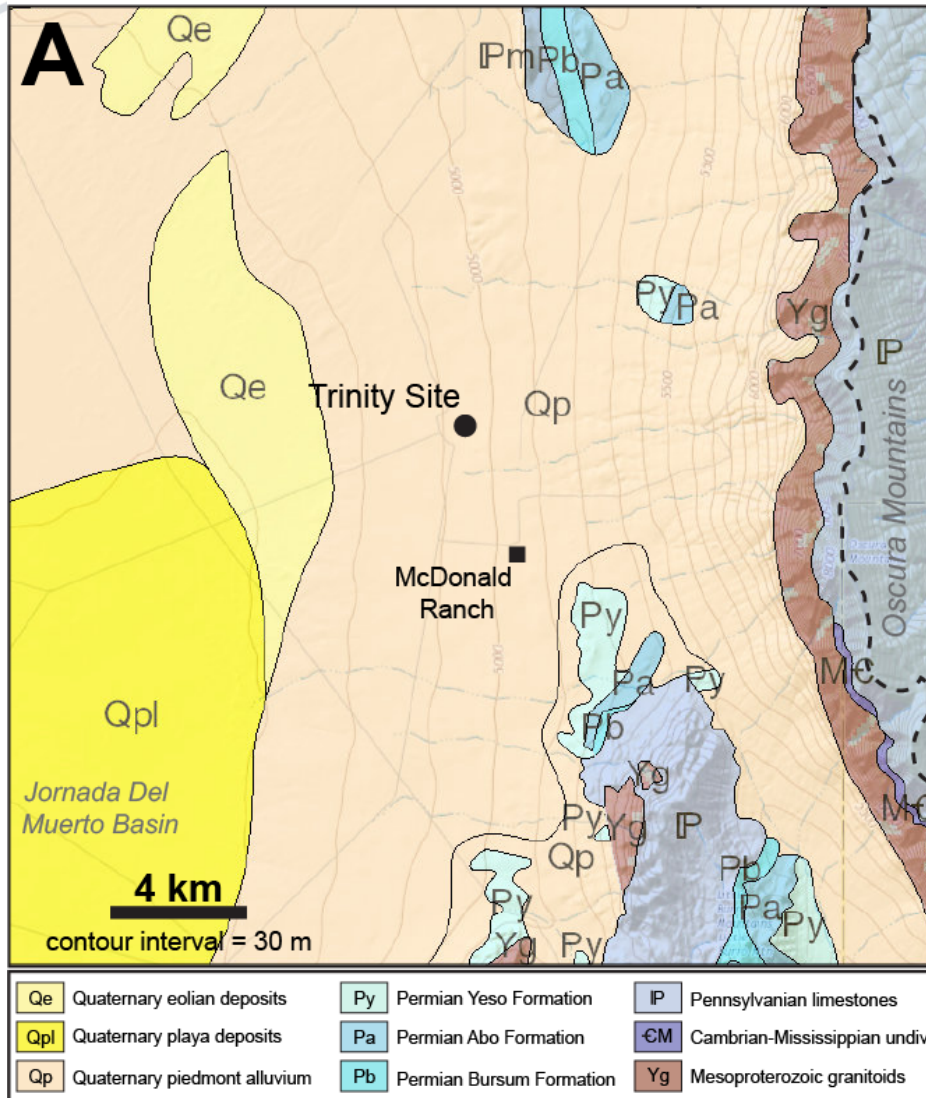
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AFTER TEST

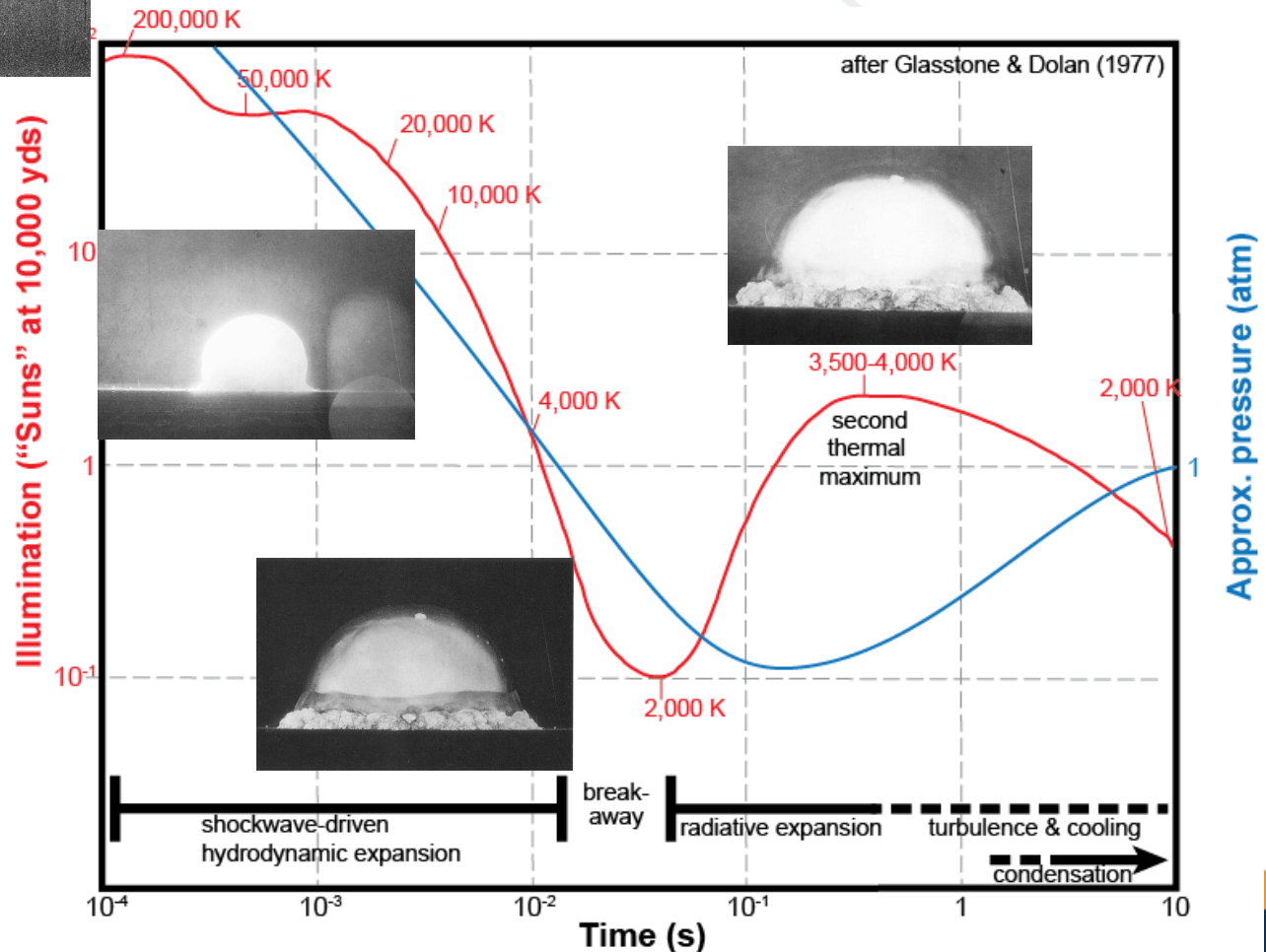
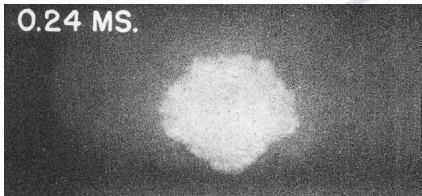
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Trinity Site Geology



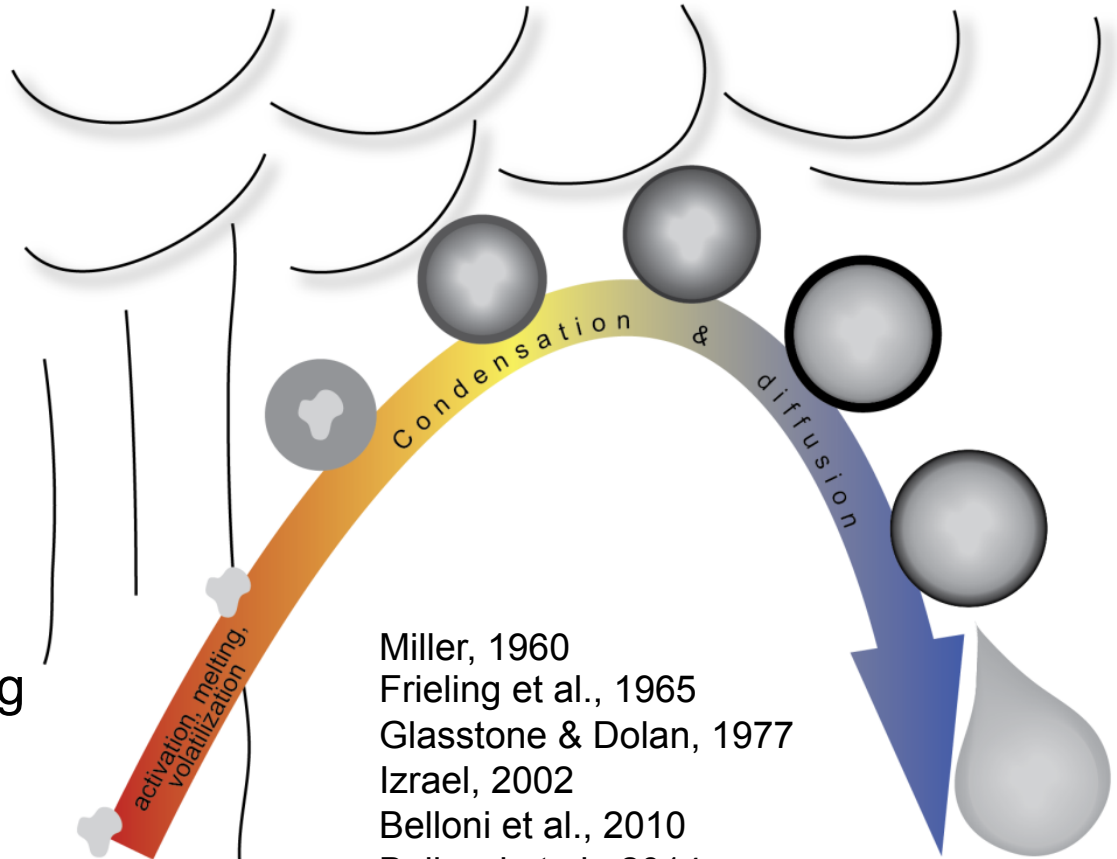
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Nuclear Fireball Phenomenology



Existing Fallout Formation Model

- Volatilization of device
- Partial melting and entrainment of ground material
- Condensation
 - Refractory bomb material
 - Ground material nuclei
- Homogenization by diffusion and/or melt mixing
- Cooling and fallout



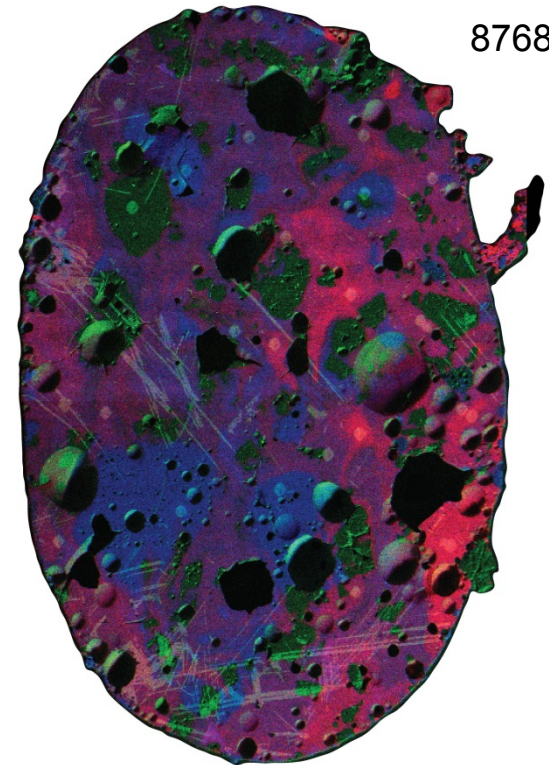
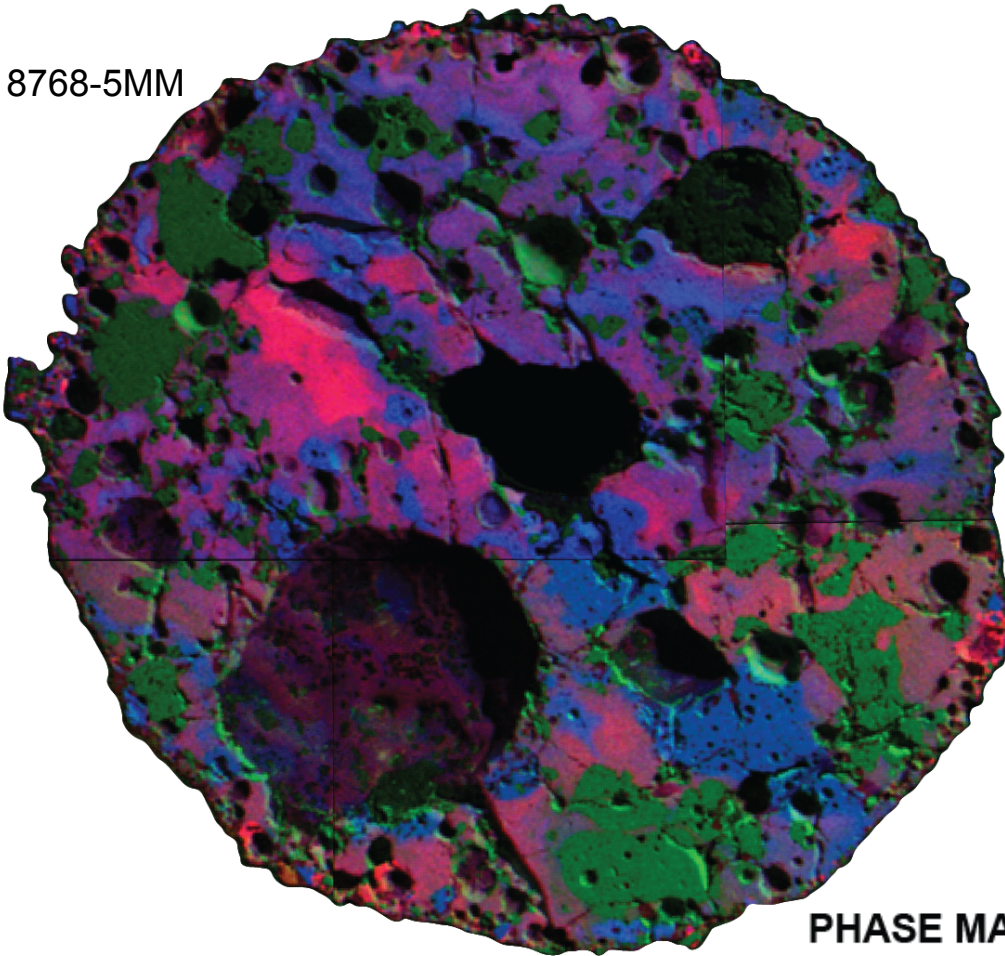
Miller, 1960
Frieling et al., 1965
Glasstone & Dolan, 1977
Izrael, 2002
Belloni et al., 2010
Bellucci et al., 2014
Eby, 2015
Lewis et al., 2015
Weisz et al., 2017

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Fallout Heterogeneity

8768-5MM

8768-4



PHASE MAPS



Quartz/
Silica glass



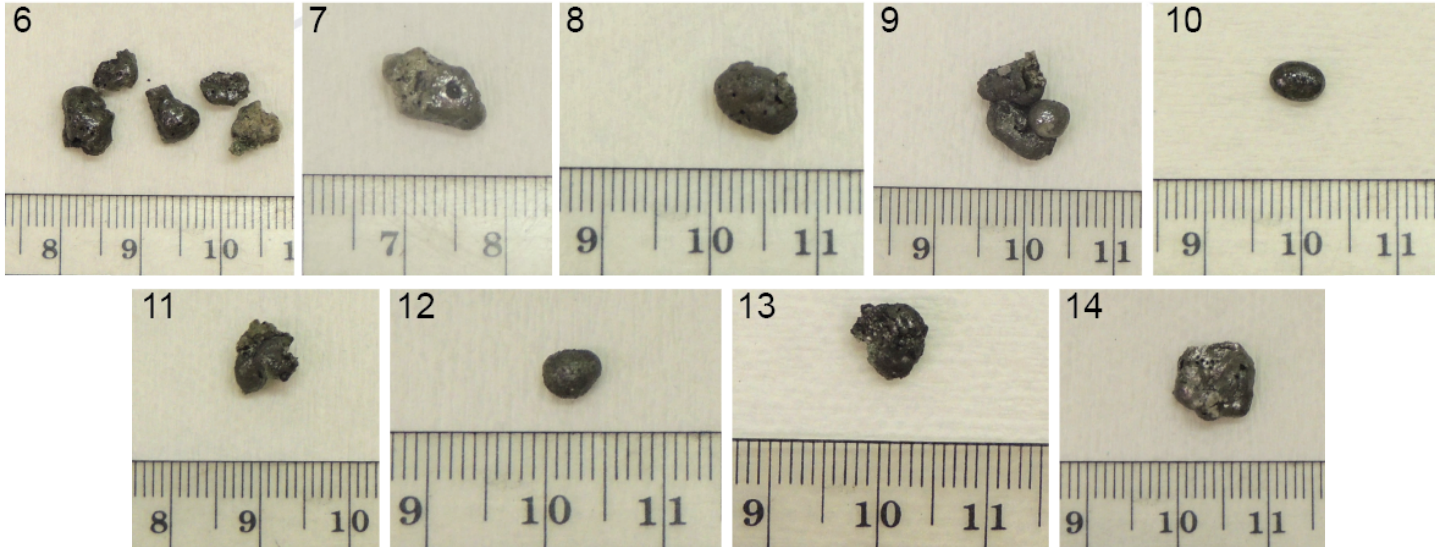
Alkali-rich glass



CaMgFe glass

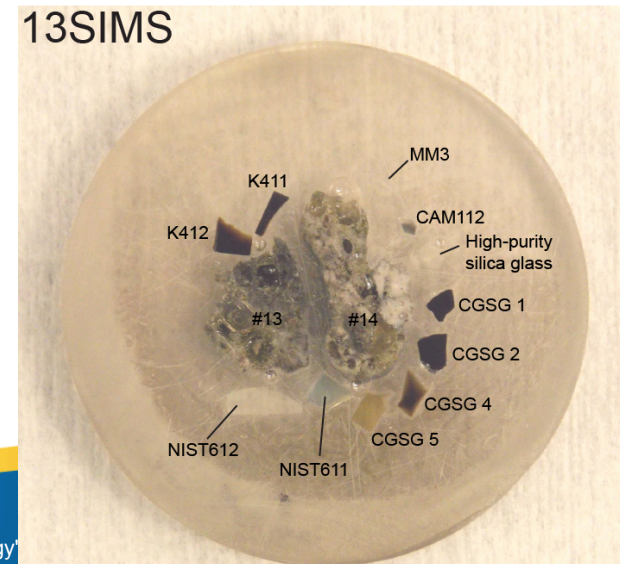
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Trinity Bead Samples

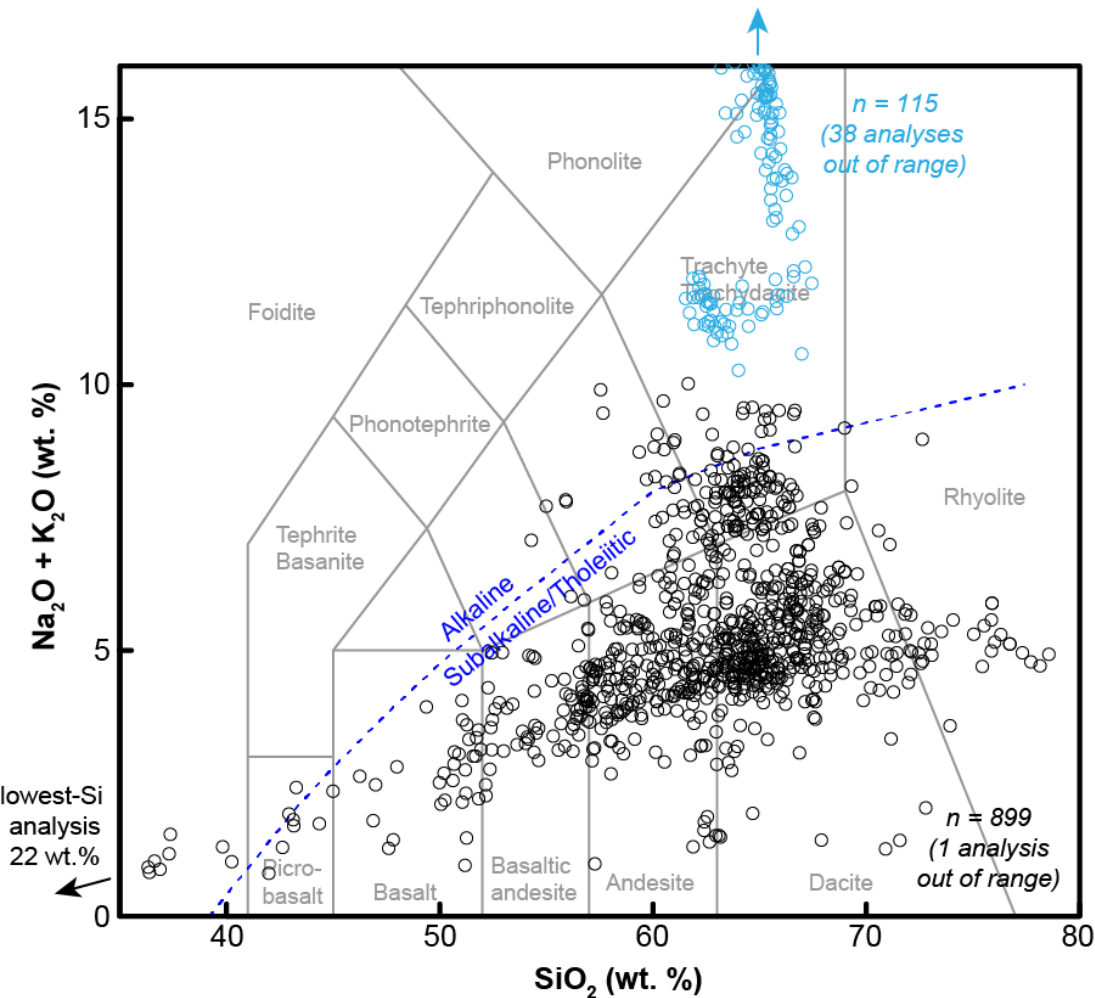


- Element mapping (SEM)
- Major Element Analysis (EPMA)
- Activity mapping (autoradiography)

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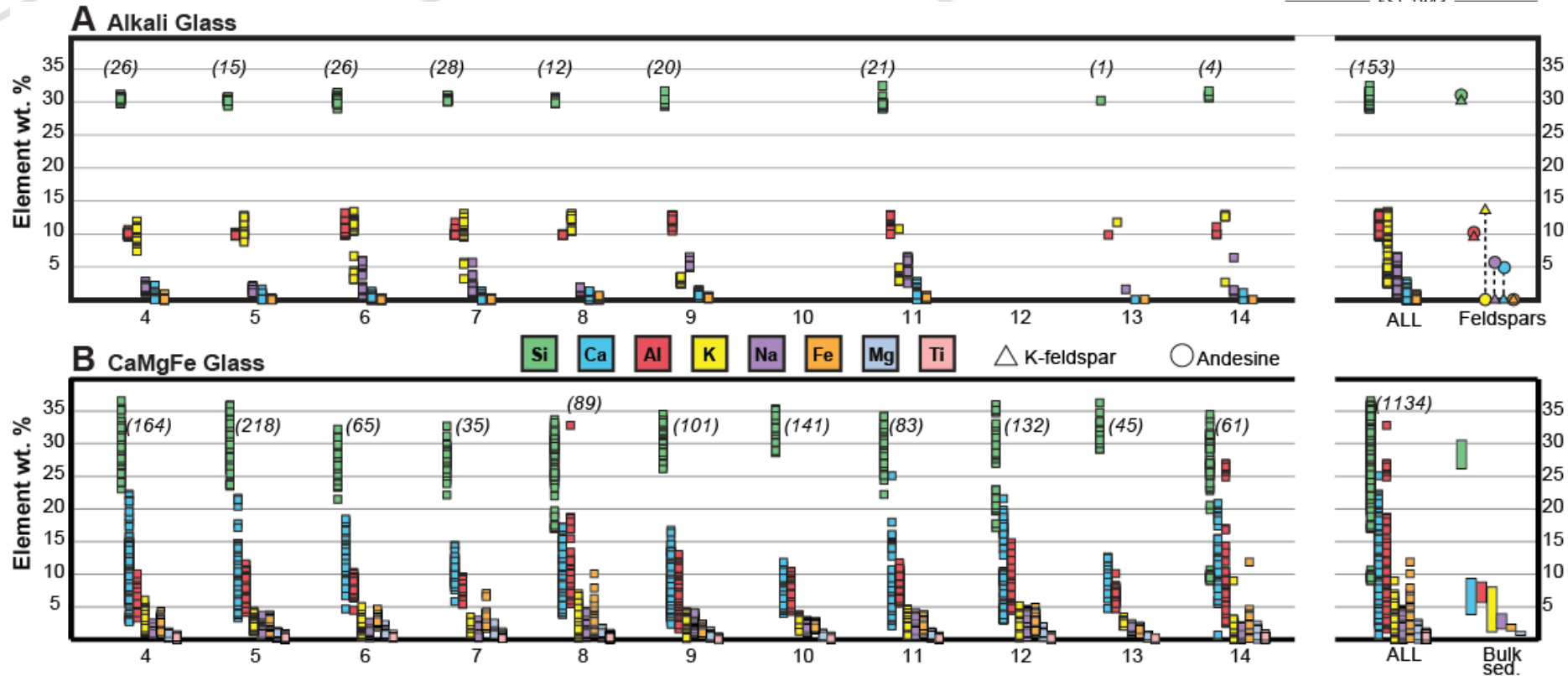
Major Element Geochemistry



- Total alkali silica diagram
- Compositional diversity beyond known magmatic range

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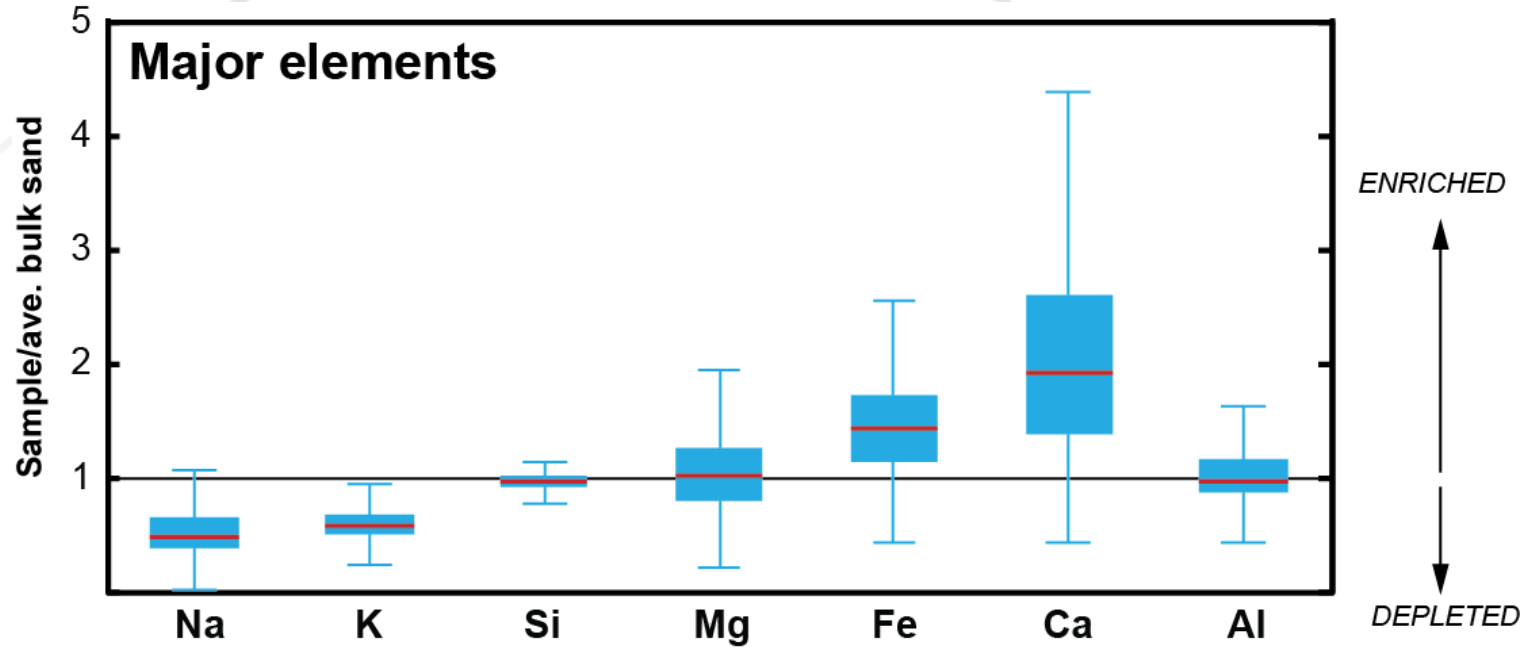
Major Element Geochemistry



- Alkali glass = melted alkali feldspar
- Extreme compositional variability in CaMgFe glass

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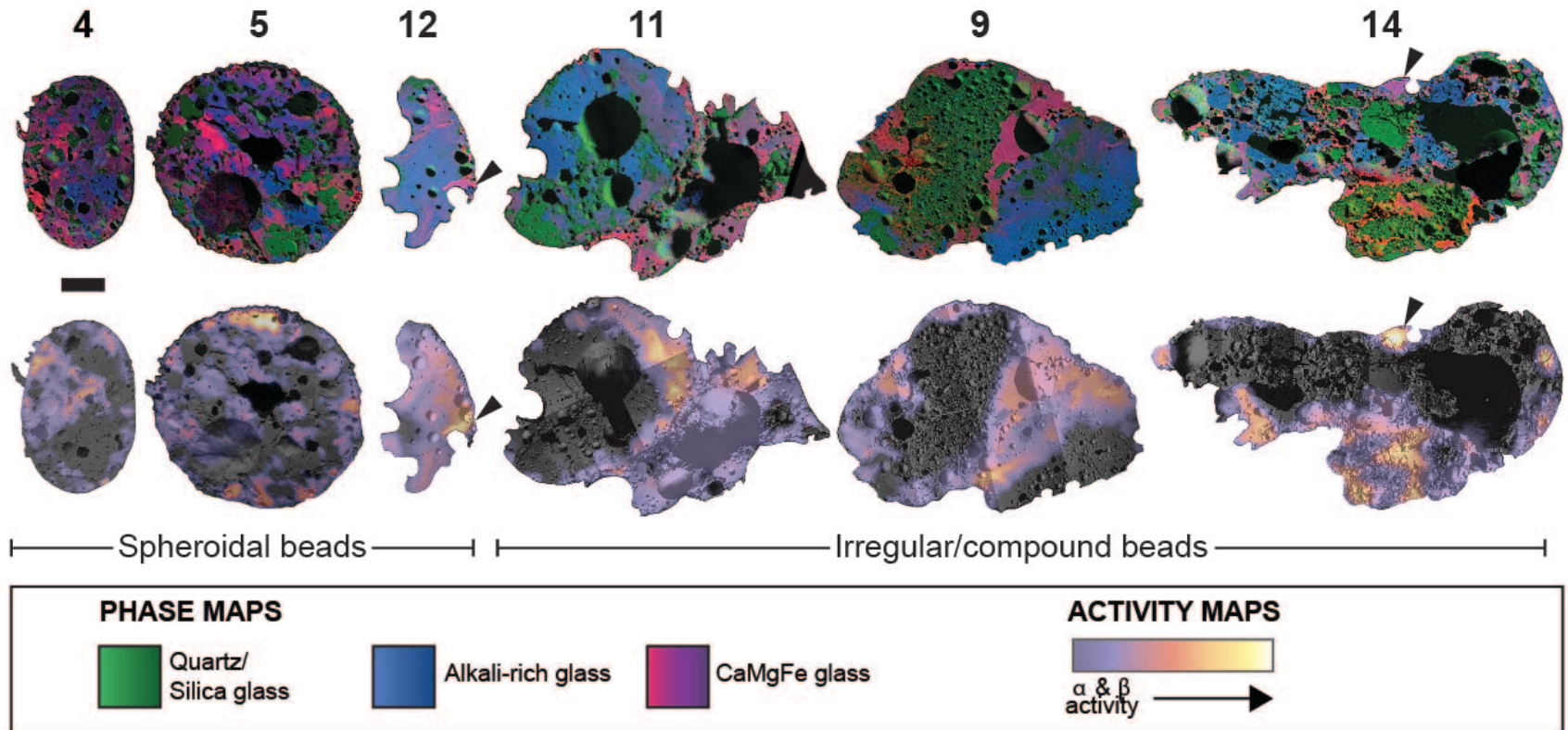
Major Element Geochemistry



- Overall depletion of volatile elements
- Enrichment in intermediate-refractory elements

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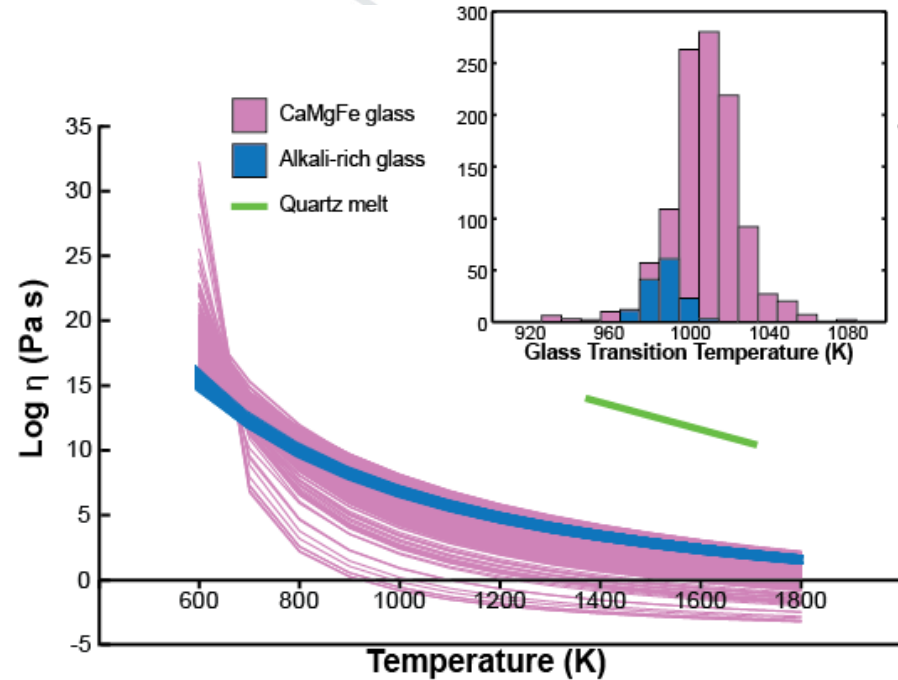
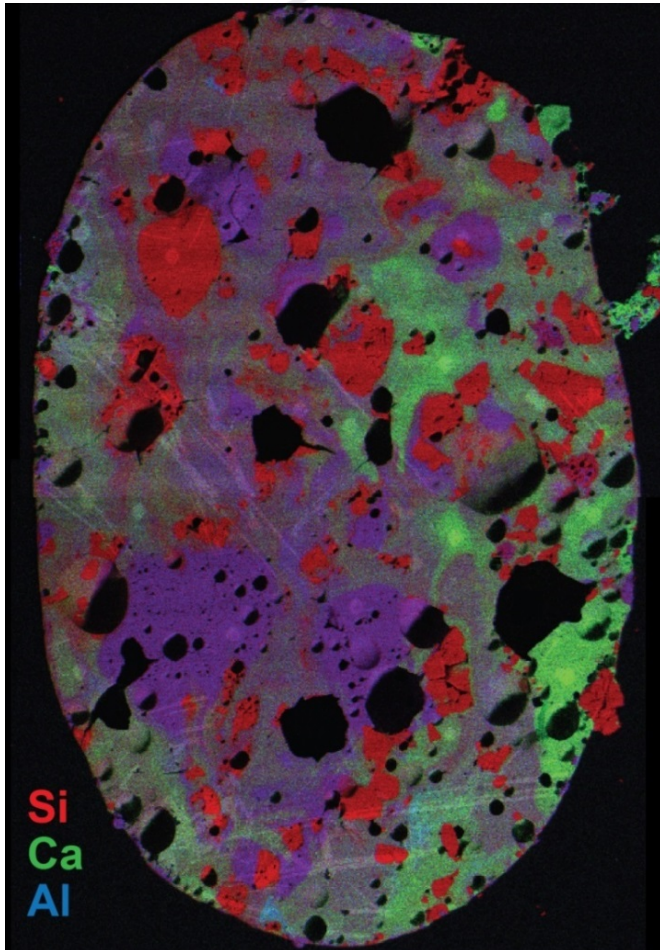
Where is the Radioactivity?



- Activity is strongly correlated with regions of CaMgFe glass, consistent with co-condensation of fissionogenic and nonfissionogenic elements

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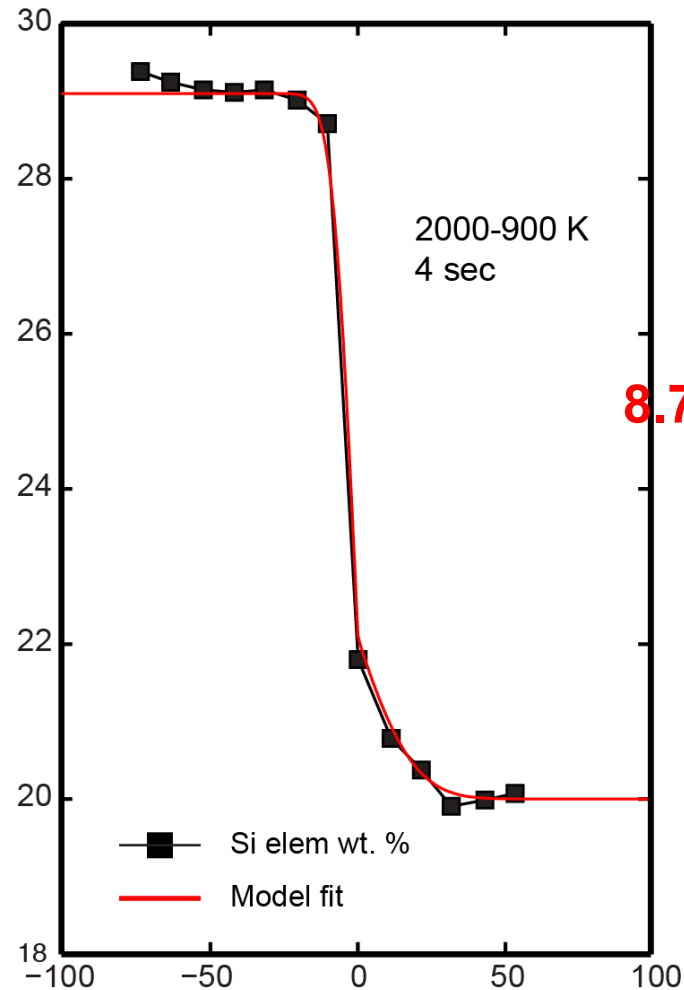
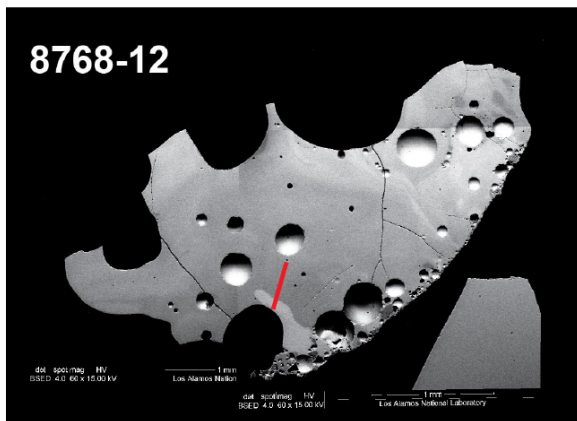
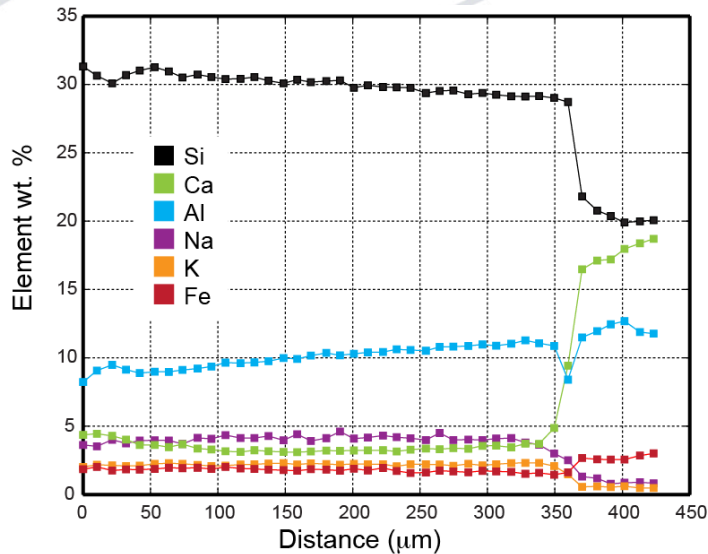
Viscosity & Melt Mixing



- Viscosity contrast likely limited physical mixing of CaMgFe, alkali, and silica melts

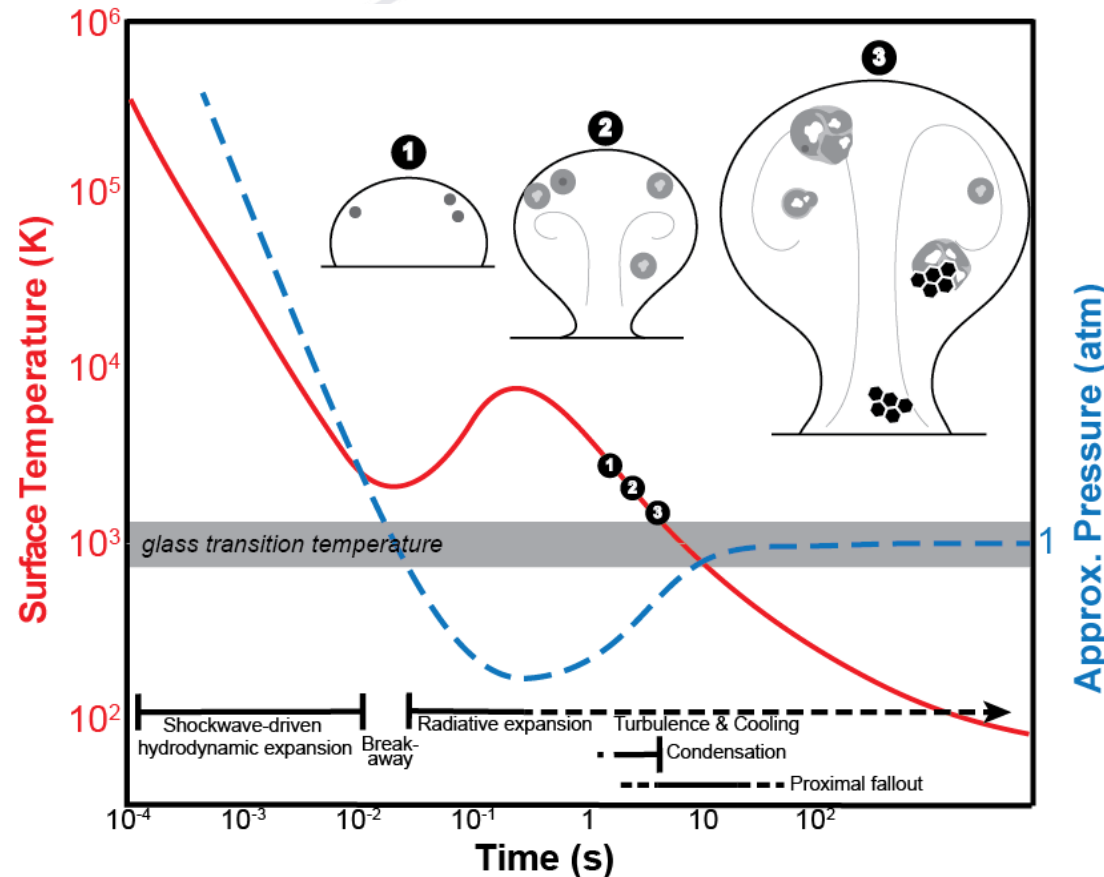
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Diffusion & Timescales of Cooling



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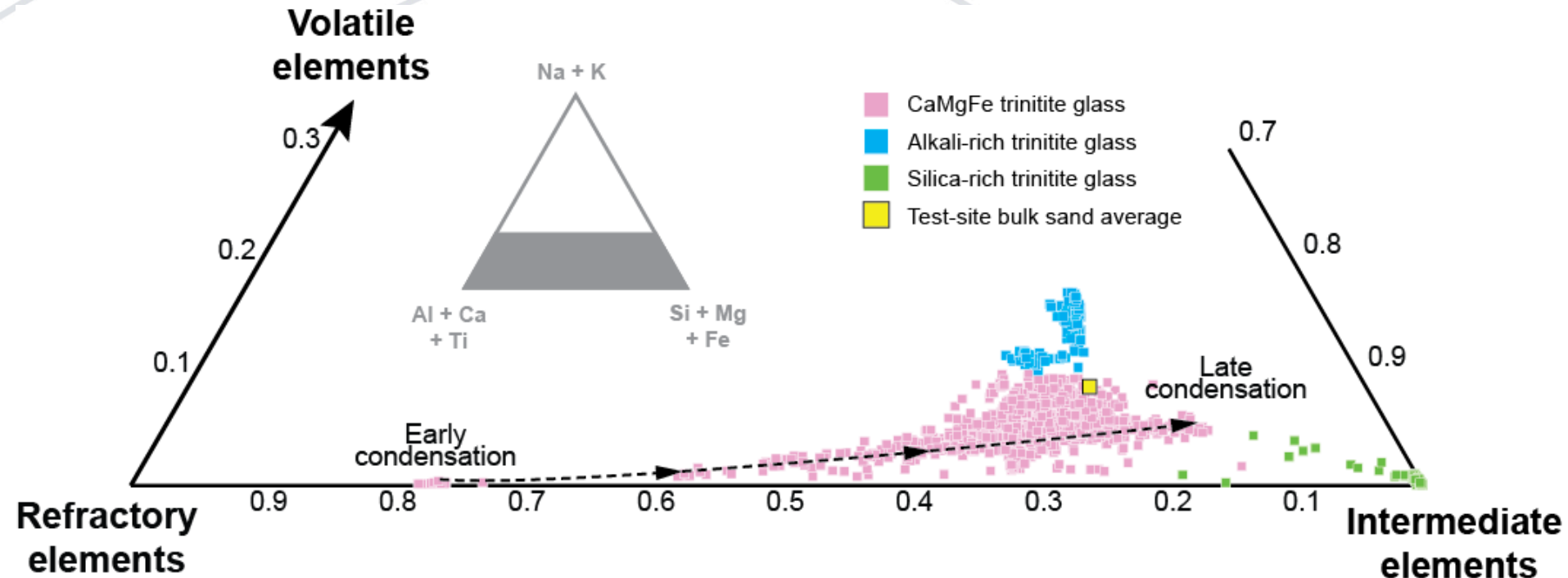
New Fallout Formation Model



- Vaporization of device, tower, sand
- Molten and solid sand entrainment
- Co-condensation of device and ground constituents
- Agglomeration
- Cooling and fallout

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Condensation Chronology & Volatility Scale

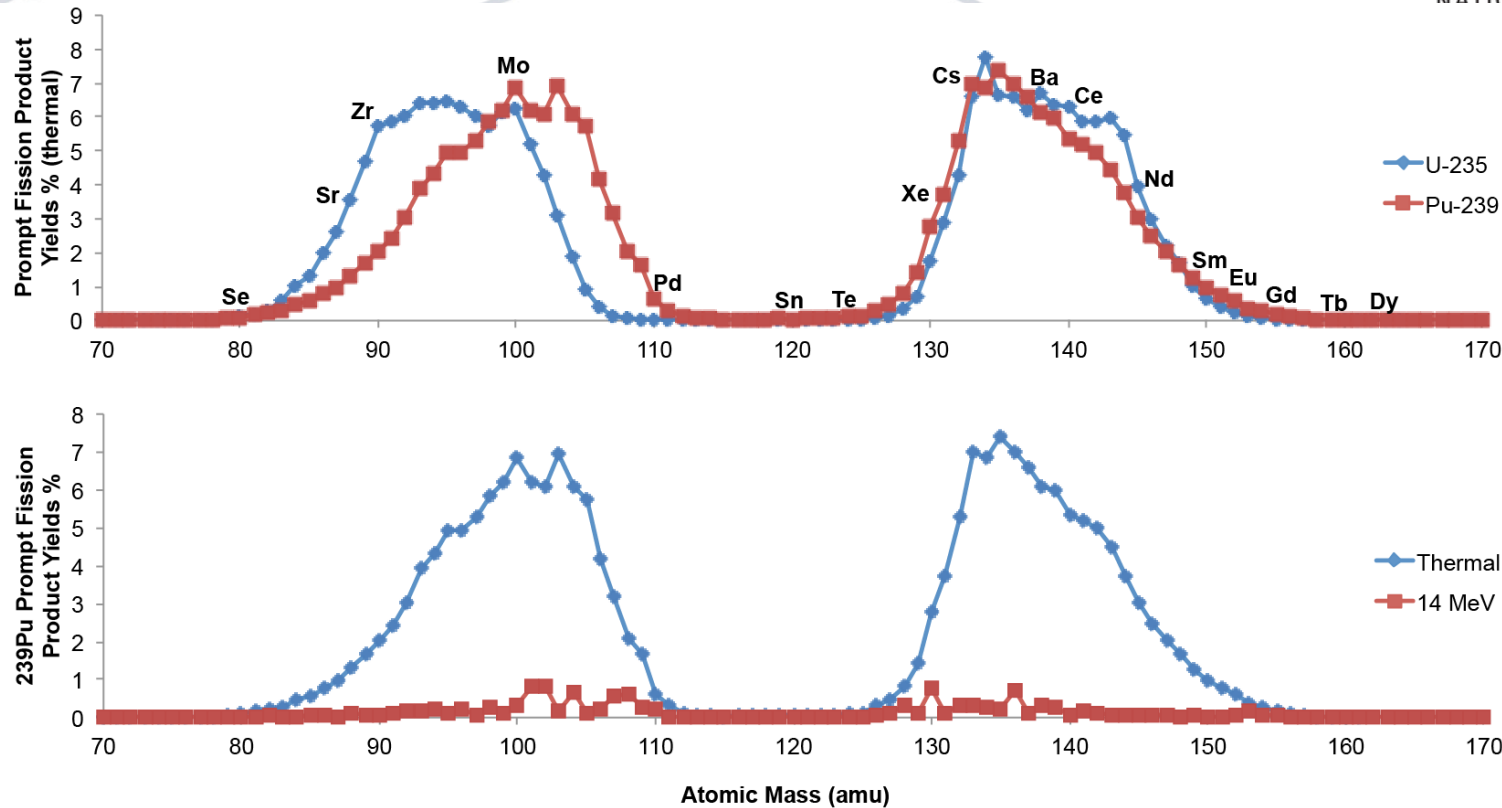


$$Volatility\ Index = \frac{volatiles}{refractories} = \frac{Na + K}{Al + Ca + Ti}$$

- Define a volatility scale based on the relative amounts of volatile and refractory major elements

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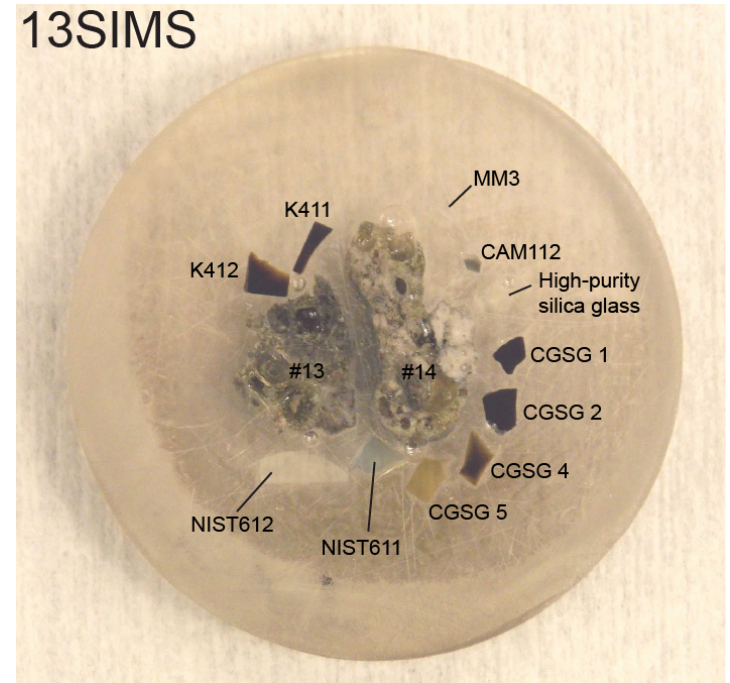
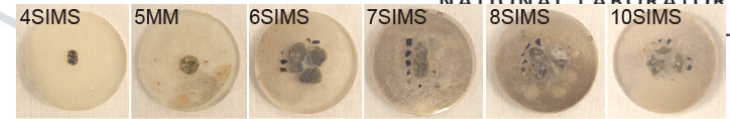
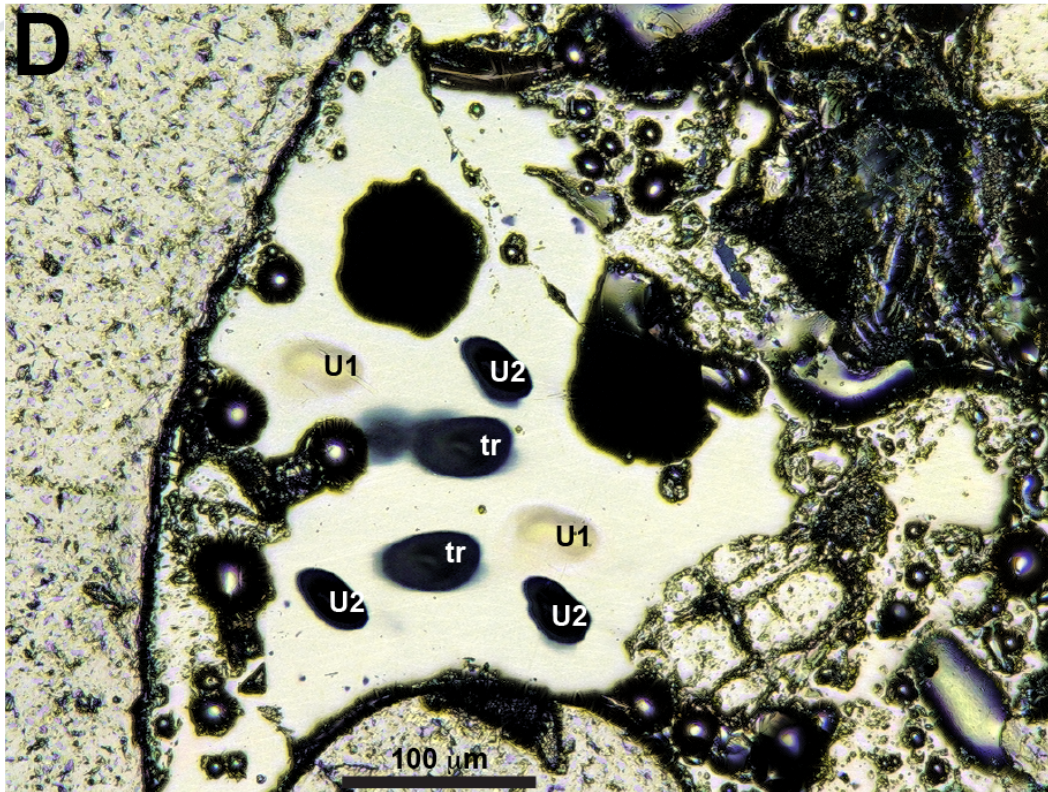
Fissionogenic Nuclides



- Yields of fissionogenic nuclides are related to the type of fuel used and the energy spectrum of neutrons produced

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Trace Elements & Isotopes Methods

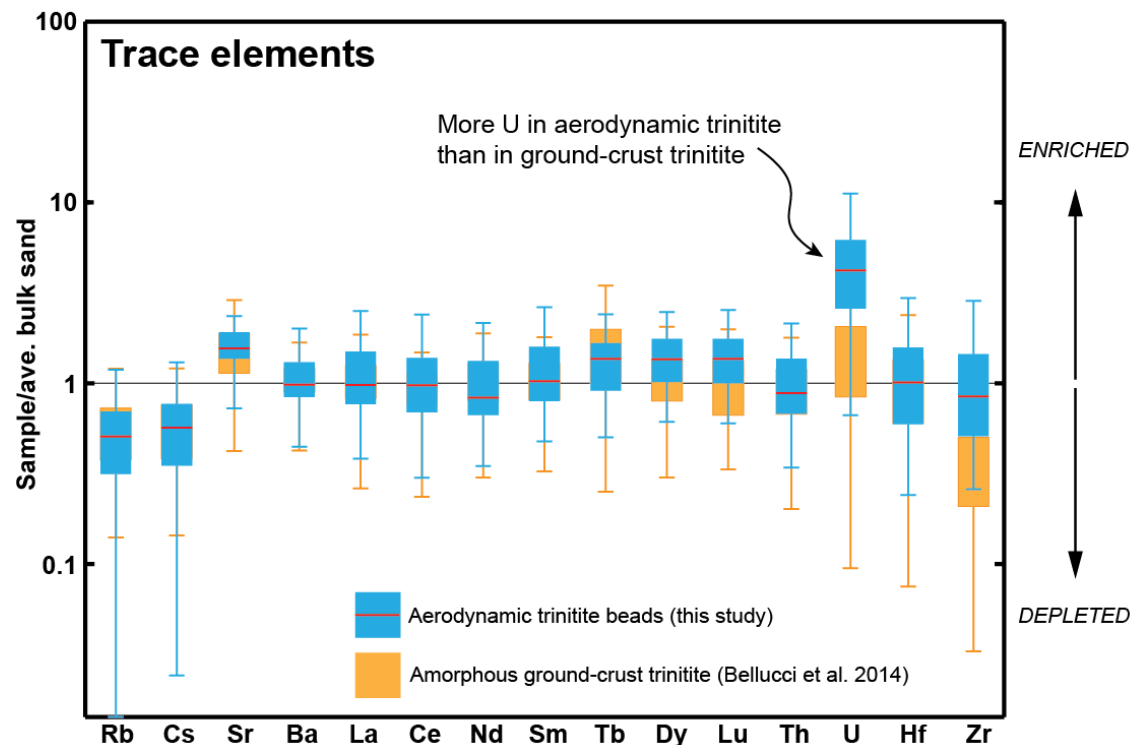
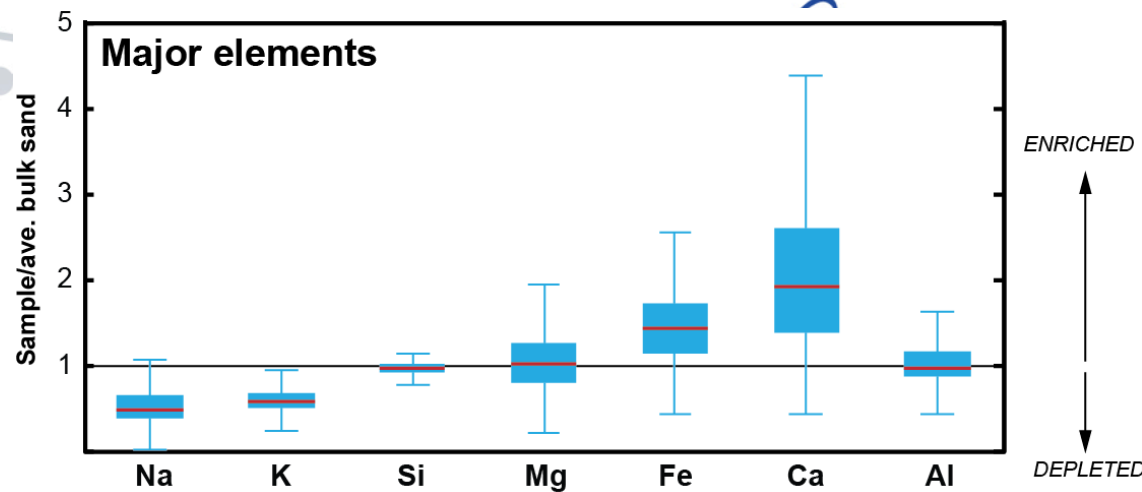


- Cameca IMS 6f SIMS for trace elements
- Cameca IMS 1280 SIMS for U-Pu isotopes

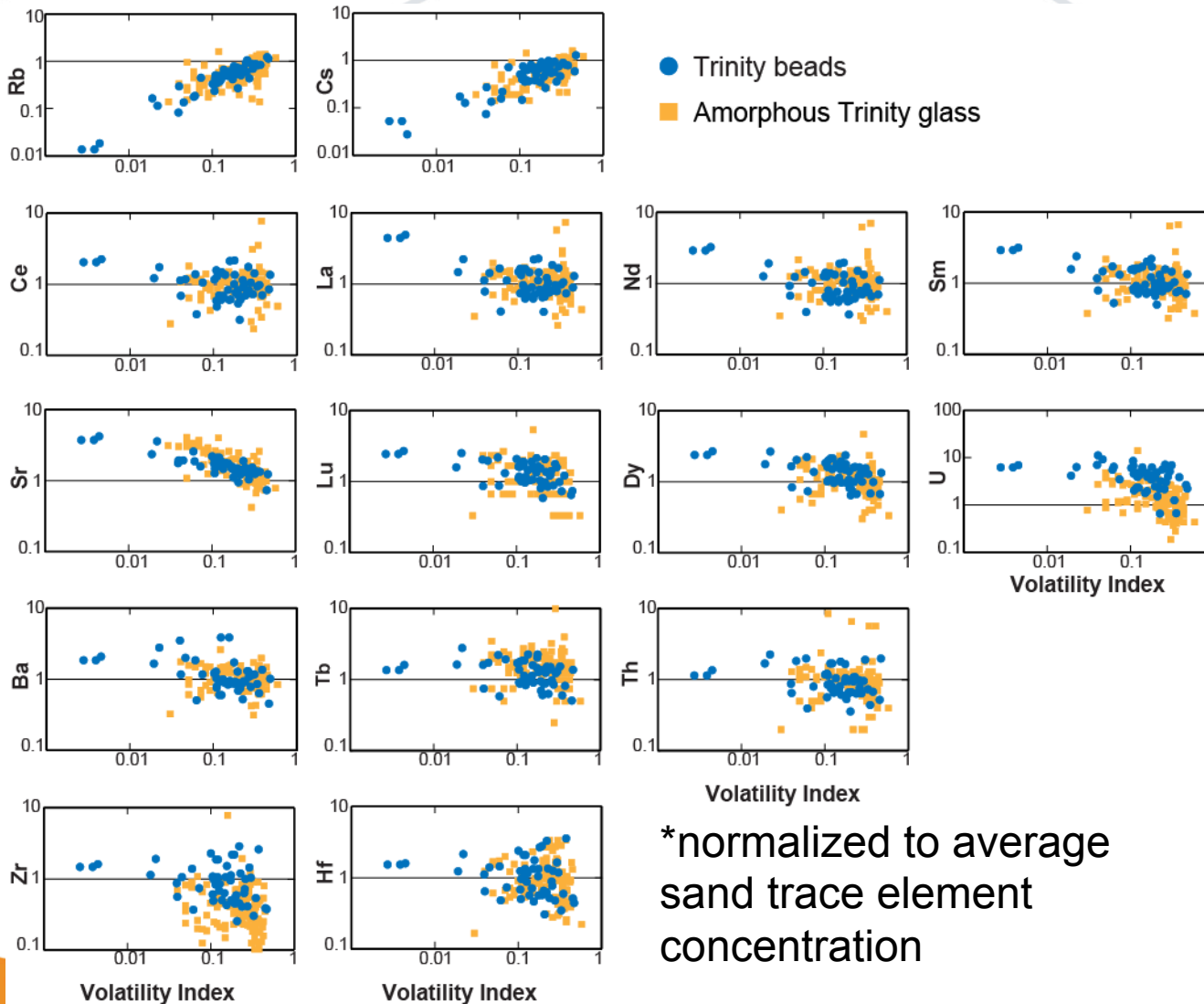
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Trace Element Geochemistry

- Similar pattern to major elements
- Enrichment of U in beads relative to ground-crust trinitite



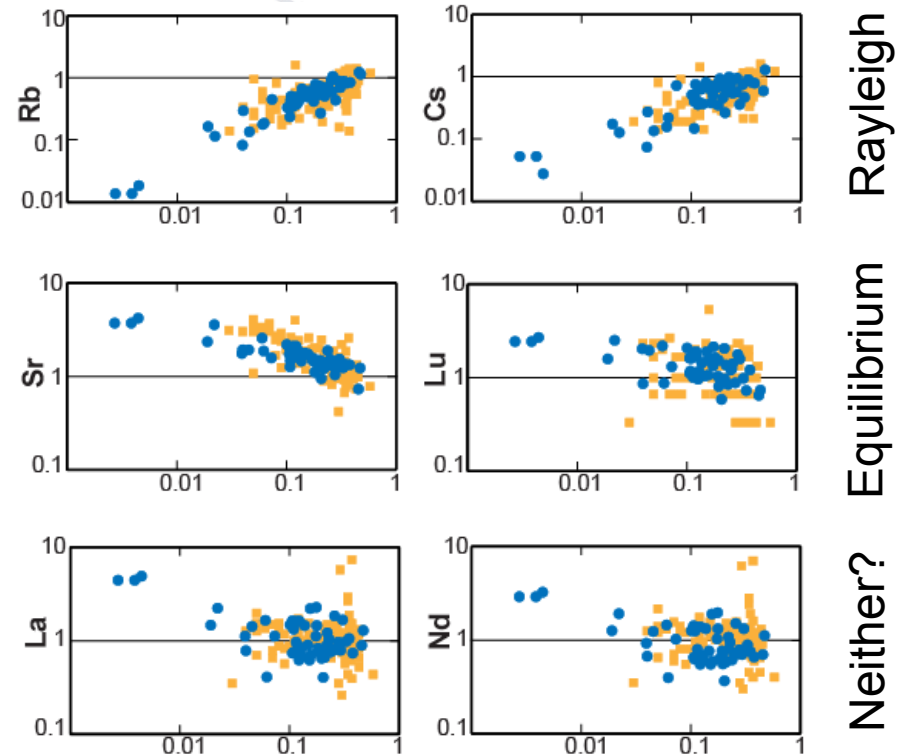
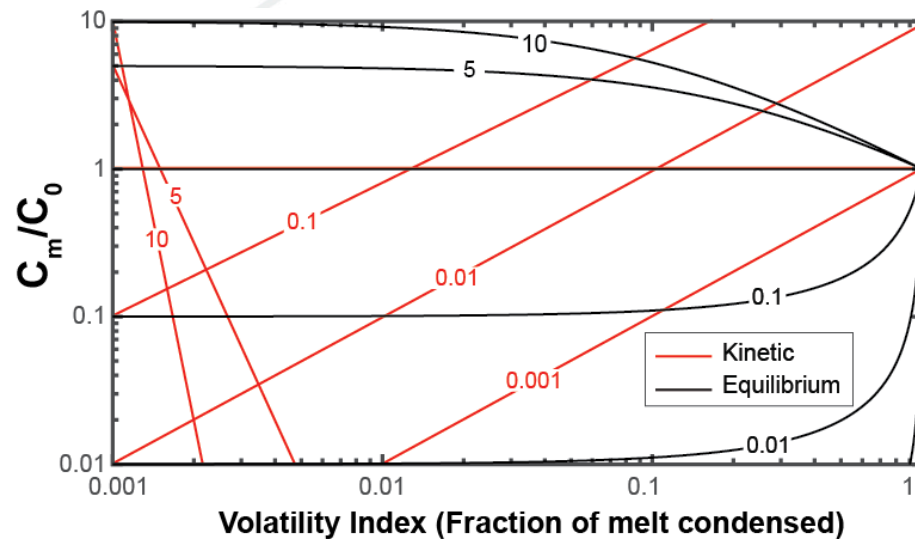
Trace Element Geochemistry



*normalized to average
sand trace element
concentration

- Coherent increasing or decreasing trends
- Greatest scatter in most refractory elements

Trace Element Geochemistry

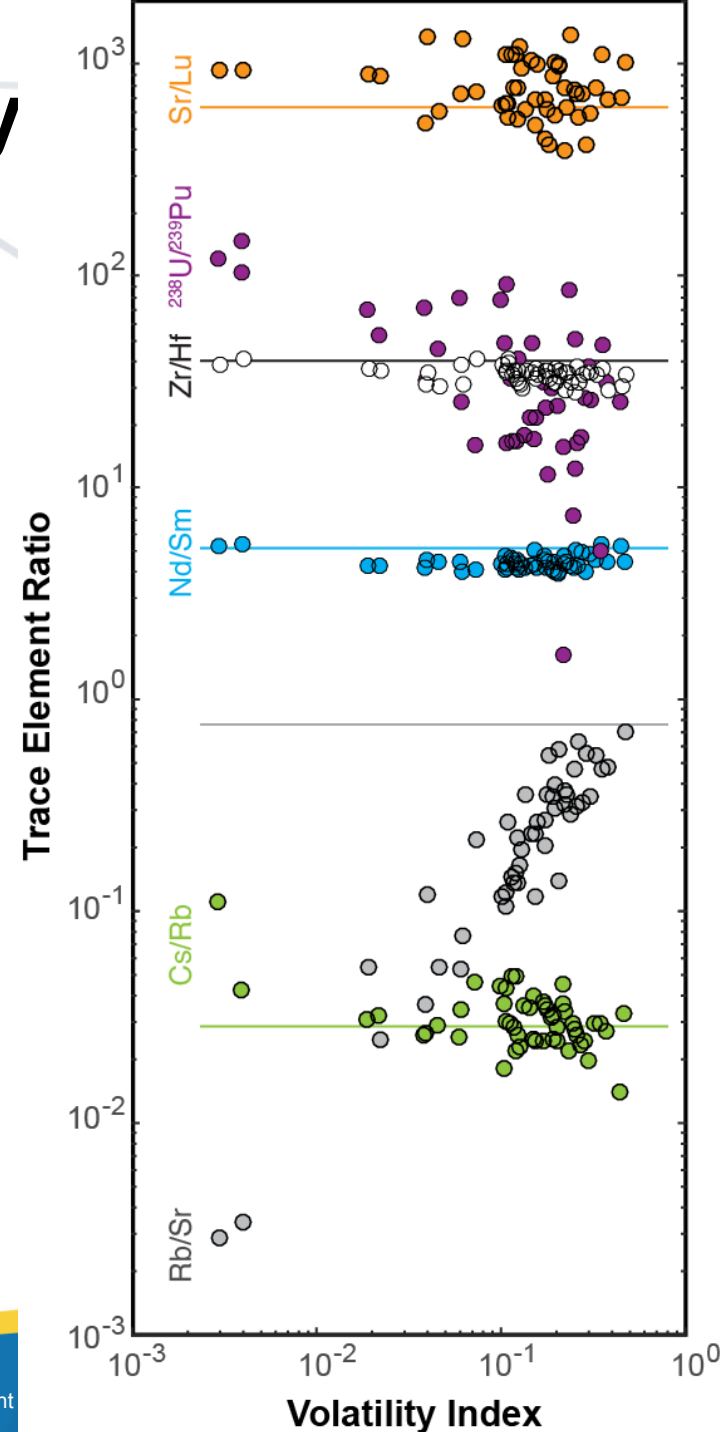


- Modeling suggests volatile elements experienced kinetic fractionation, whereas refractory elements experienced largely equilibrium fractionation

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Trace Element Geochemistry

- Track element fractionation through ratios as a function of volatility index
- Strong element fractionations → sloped trends (e.g., Rb/Sr)
- Weak element fractionations → flat trends (e.g., Zr/Hf)



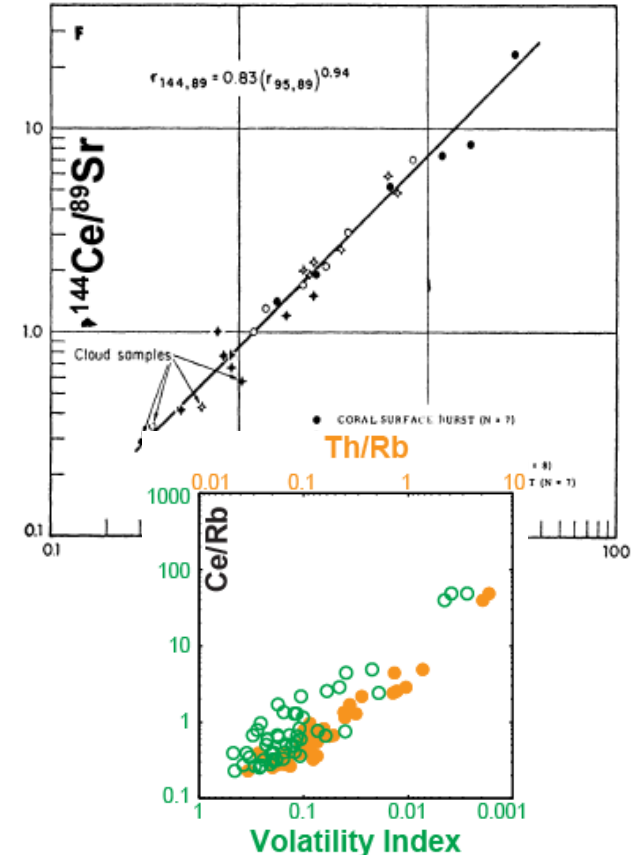
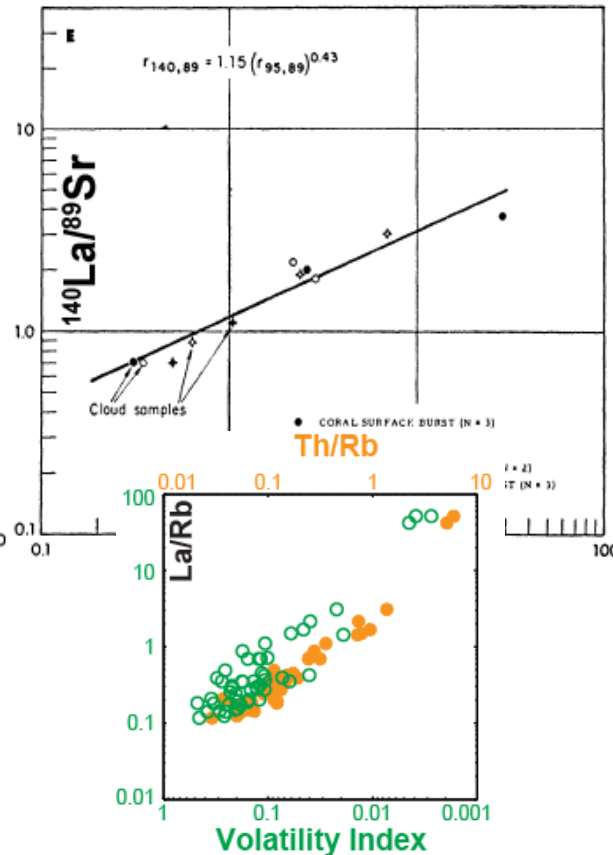
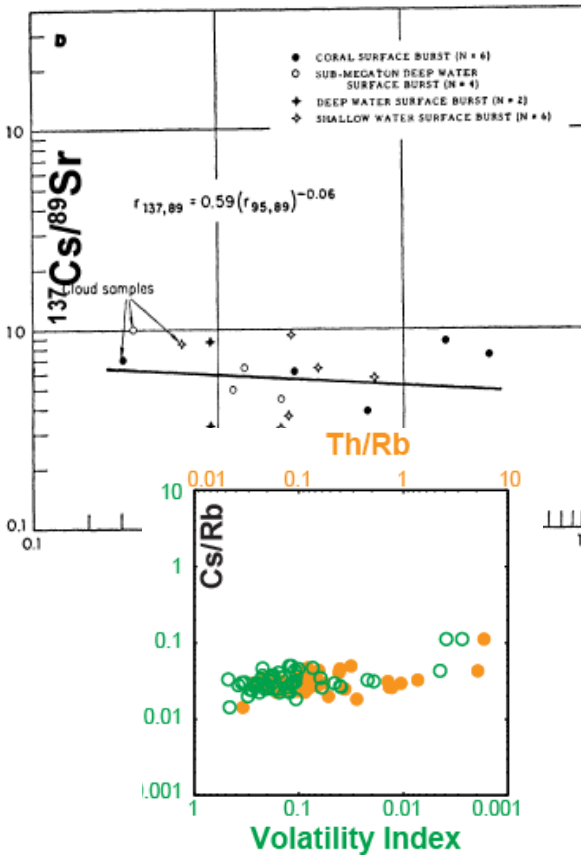
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Comparison with Freiling (1961)

$^{95}\text{Zr}/^{89}\text{Sr}$

$^{95}\text{Zr}/^{89}\text{Sr}$

$^{95}\text{Zr}/^{89}\text{Sr}$

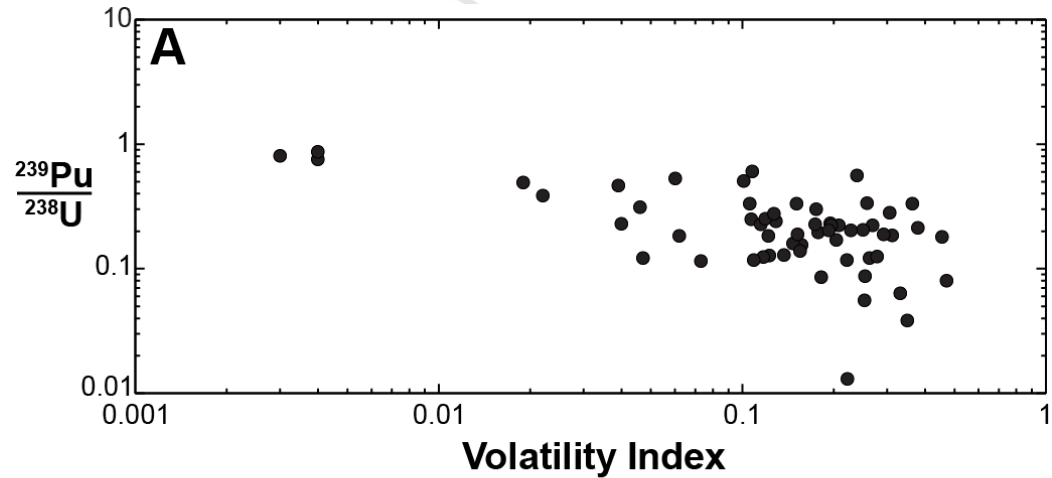


- Stable element-based volatility scales give similar results to fissionogenic isotope-based volatility scales

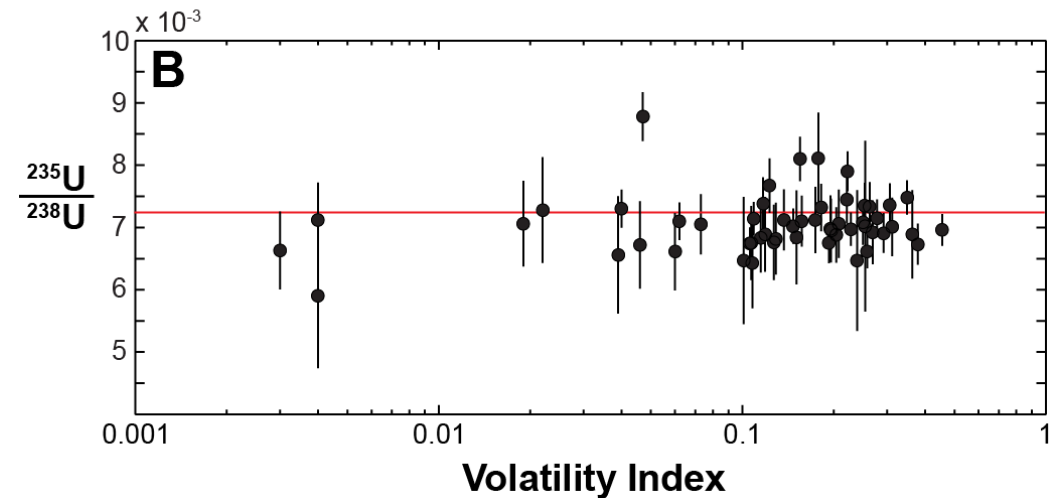
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U-Pu Isotope Geochemistry

- Pu and U fractionated significantly in the Trinity fireball



- U isotopes did not fractionate in the Trinity fireball



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Nuclear Forensics Summary

- Test-site sand vaporized and co-condensed along with radioactive bomb material
- Glassy fallout preserves compositional variations of the nuclear fireball
- Stable elements are appropriate proxies for studying radionuclide fractionation during nuclear explosions
- Heavy isotopes were not fractionated during the Trinity explosion

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Natural Glasses

Impactites



gia.edu

Discovery Channel

Fulgurites



Volcanic glasses

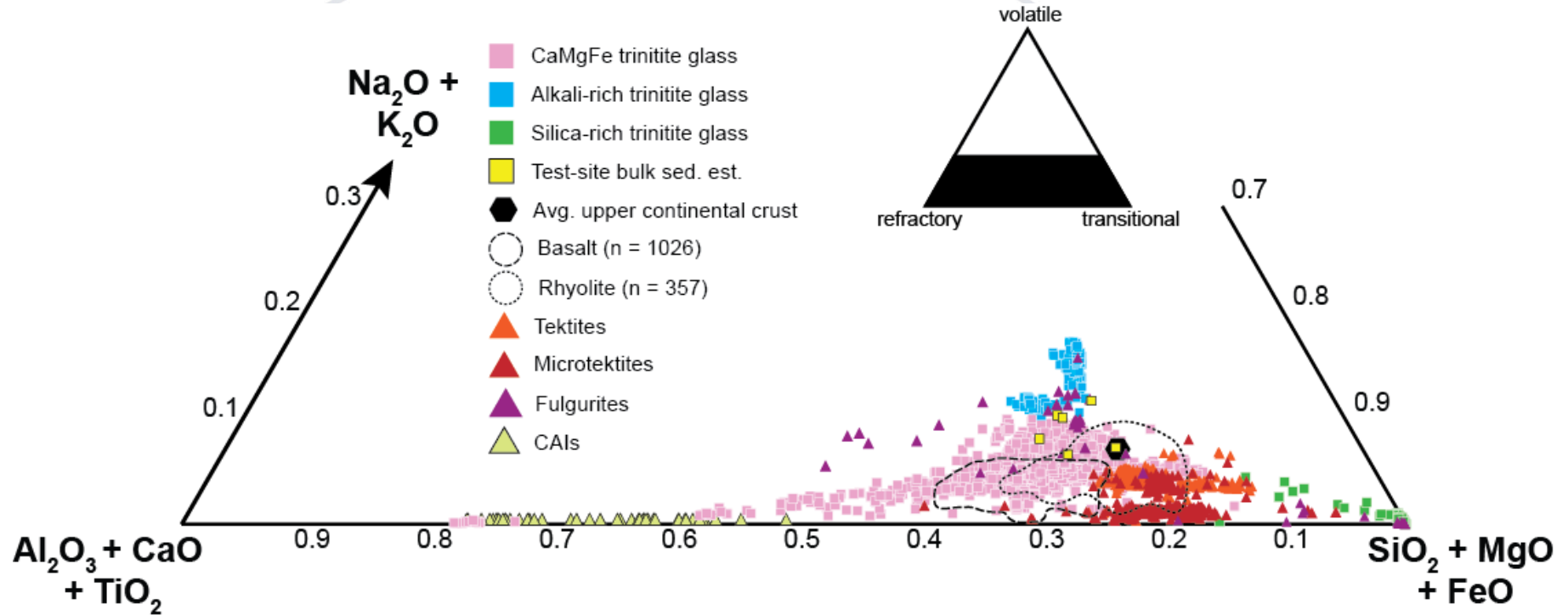


Michael Manga



nps.gov

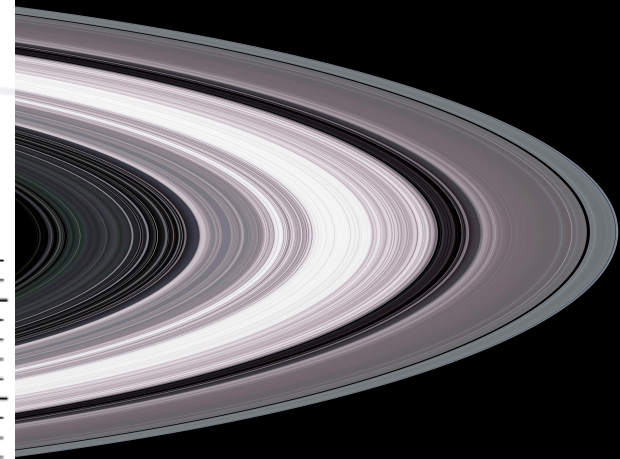
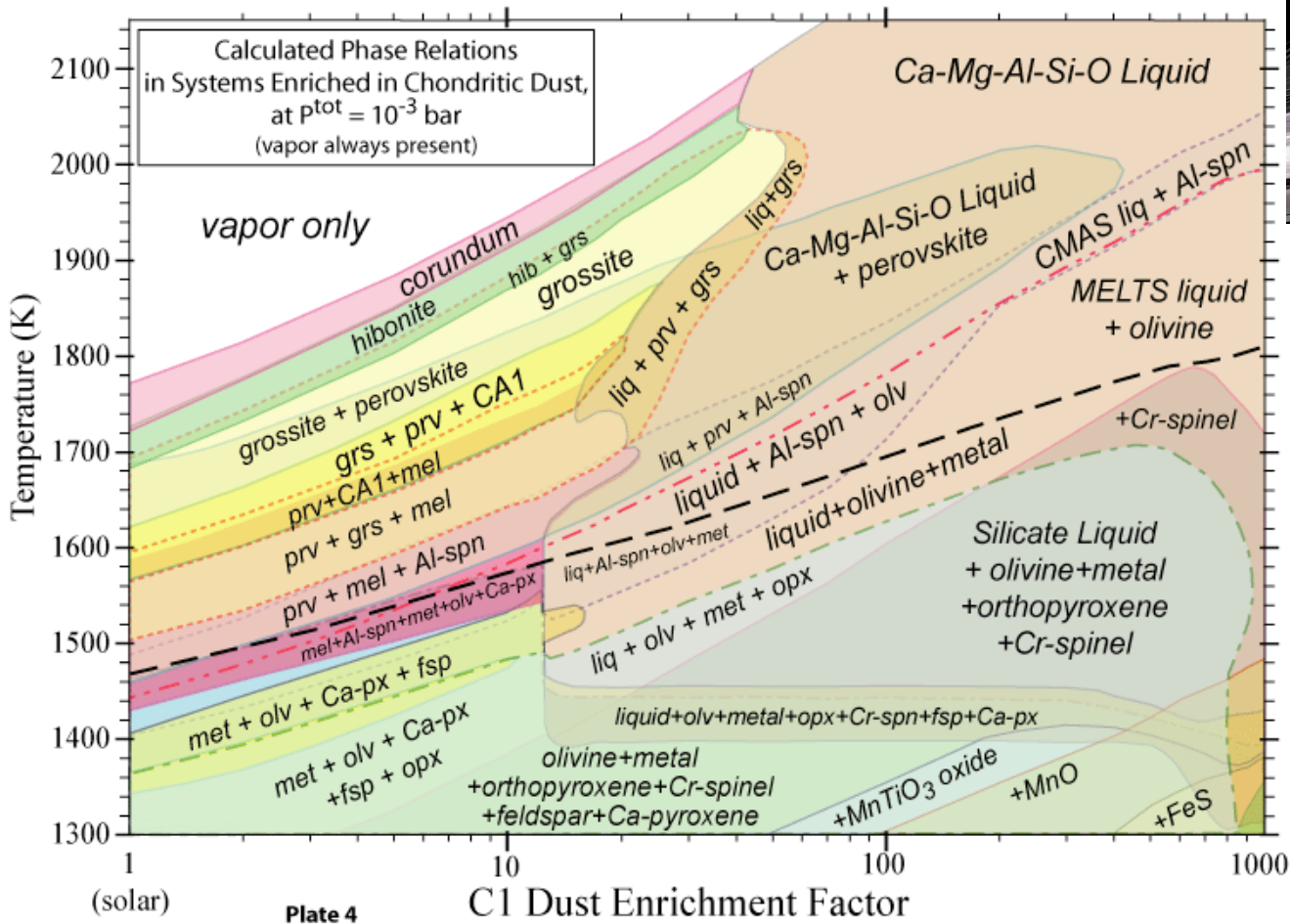
Natural Glasses



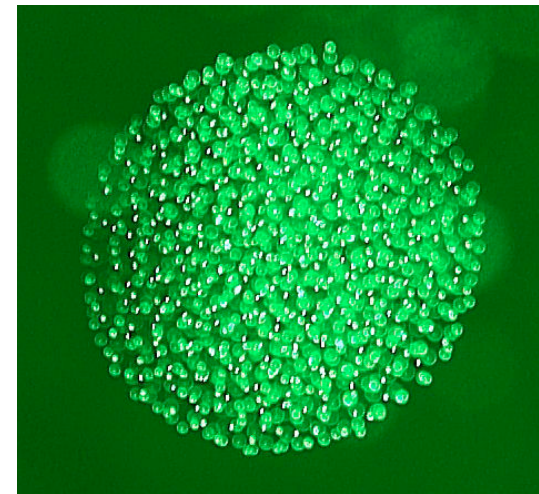
- Naturally formed glasses show a more limited compositional range than trinitite

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Dusty Plasmas



NASA/JPL



Yukawa ball
Kading & Melzer 2006

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Geochemistry Summary

- Naturally formed glasses show a more limited compositional range than trinitite
- Melting and vaporization behavior of natural glasses differed from those of trinitite, reflecting more heterogeneous energy distributions or greater physical fractionation
- Highly refractory fallout glass may be a process analogue for early solar system condensates like CAIs

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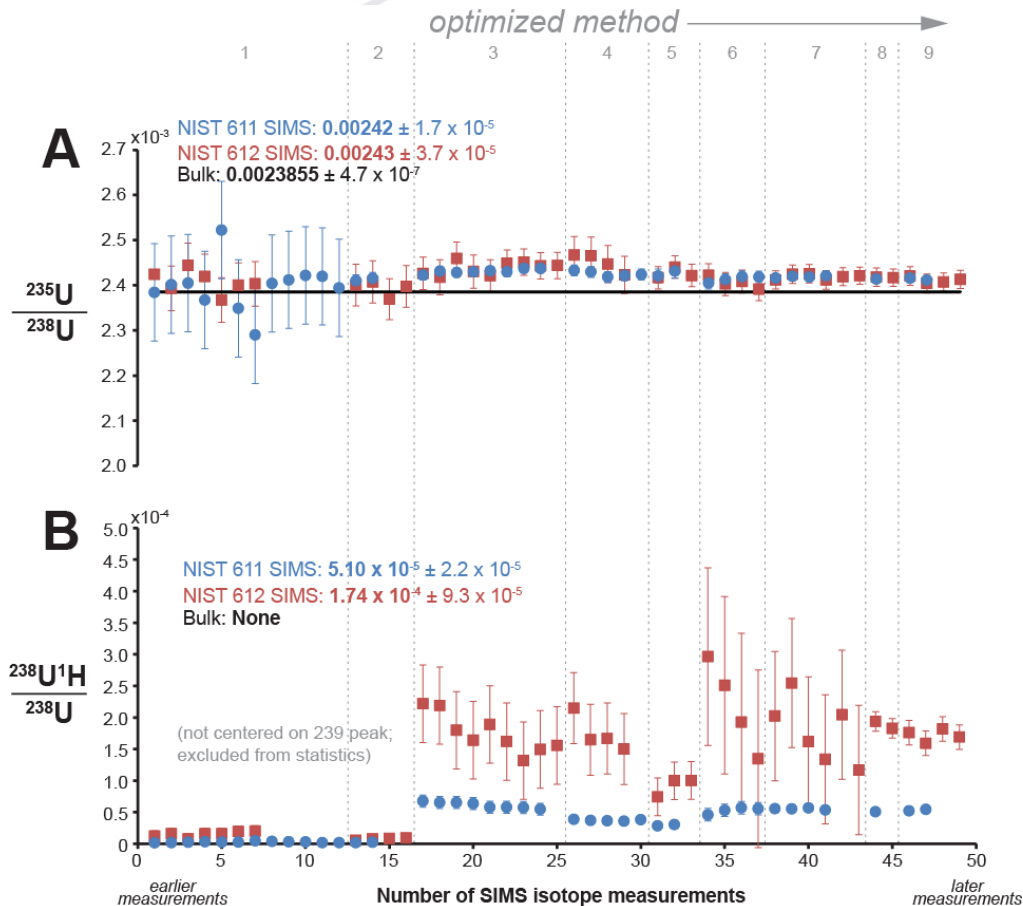
Questions?



- This work was supported by:
 - A Los Alamos National Lab Seaborg-Strategic Outcomes Office Postdoctoral Fellowship to C. Bonamici
 - Office of Defense Nuclear Nonproliferation R&D Program (NA-22), through W. Kinman

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U-Pu Isotope Measurements

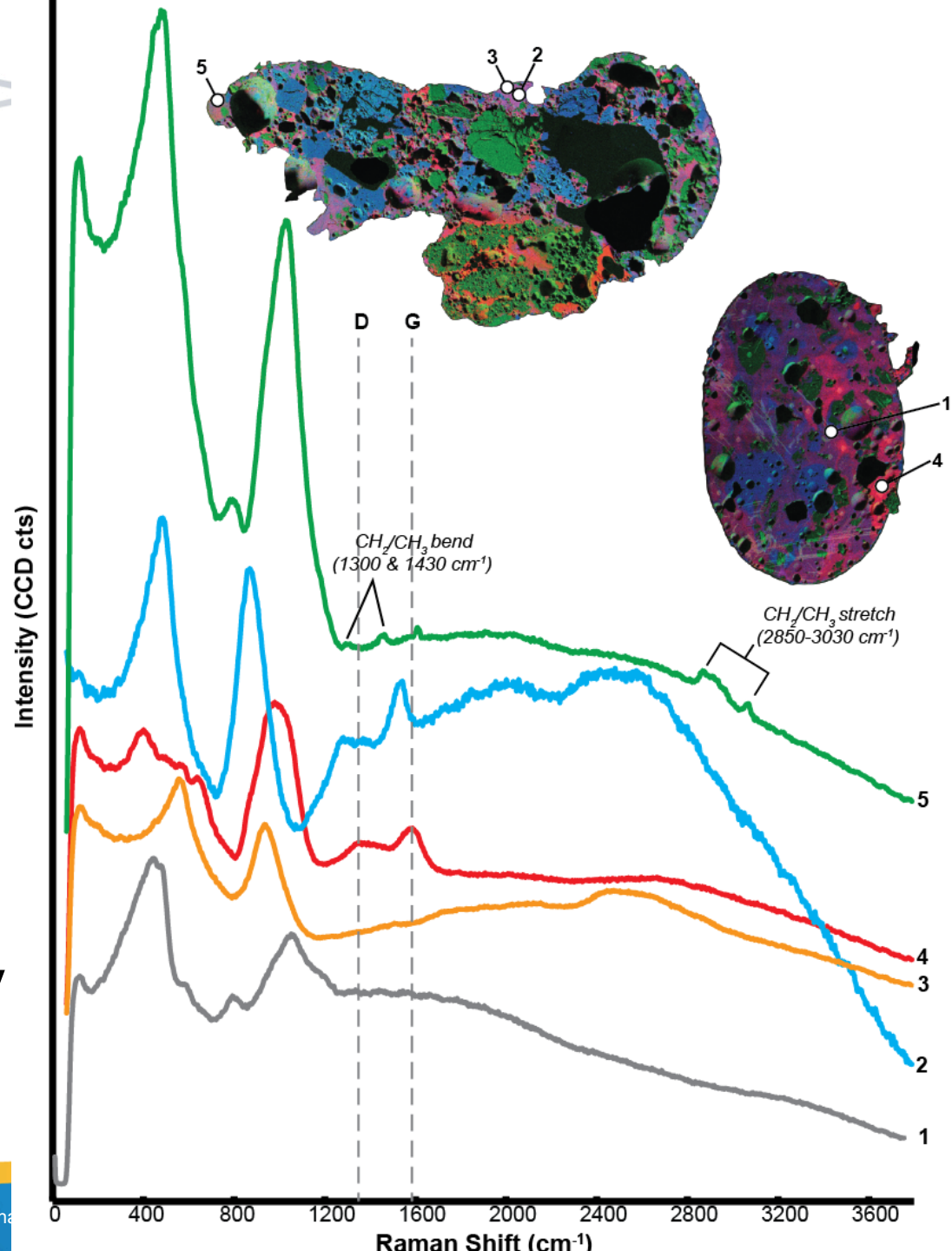


- Cameca IMS 1280
 - Static multicollector
 - ^{234}U , ^{235}U , ^{236}U , ^{238}U , ^{239}Pu
 - O⁻ primary beam
 - 420 cycles, 32 min
 - 2200 MRP
- NIST 611, NIST 612
- Hydride correction for $^{238}\text{U}^1\text{H}$ on ^{239}Pu
- Ingrowth correction for $^{239}\text{Pu} \rightarrow ^{235}\text{U}$

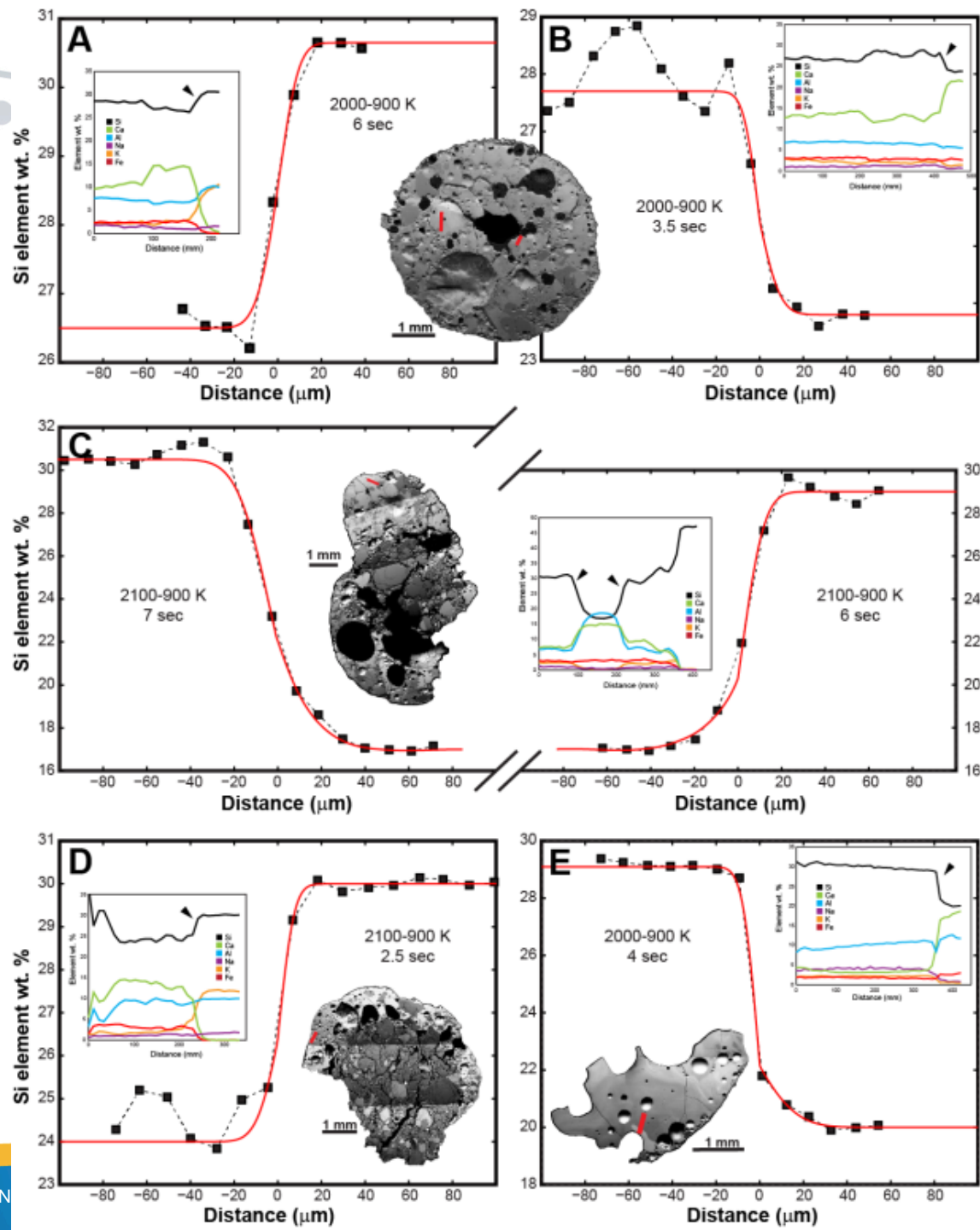
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Oxidation State in the Fireball

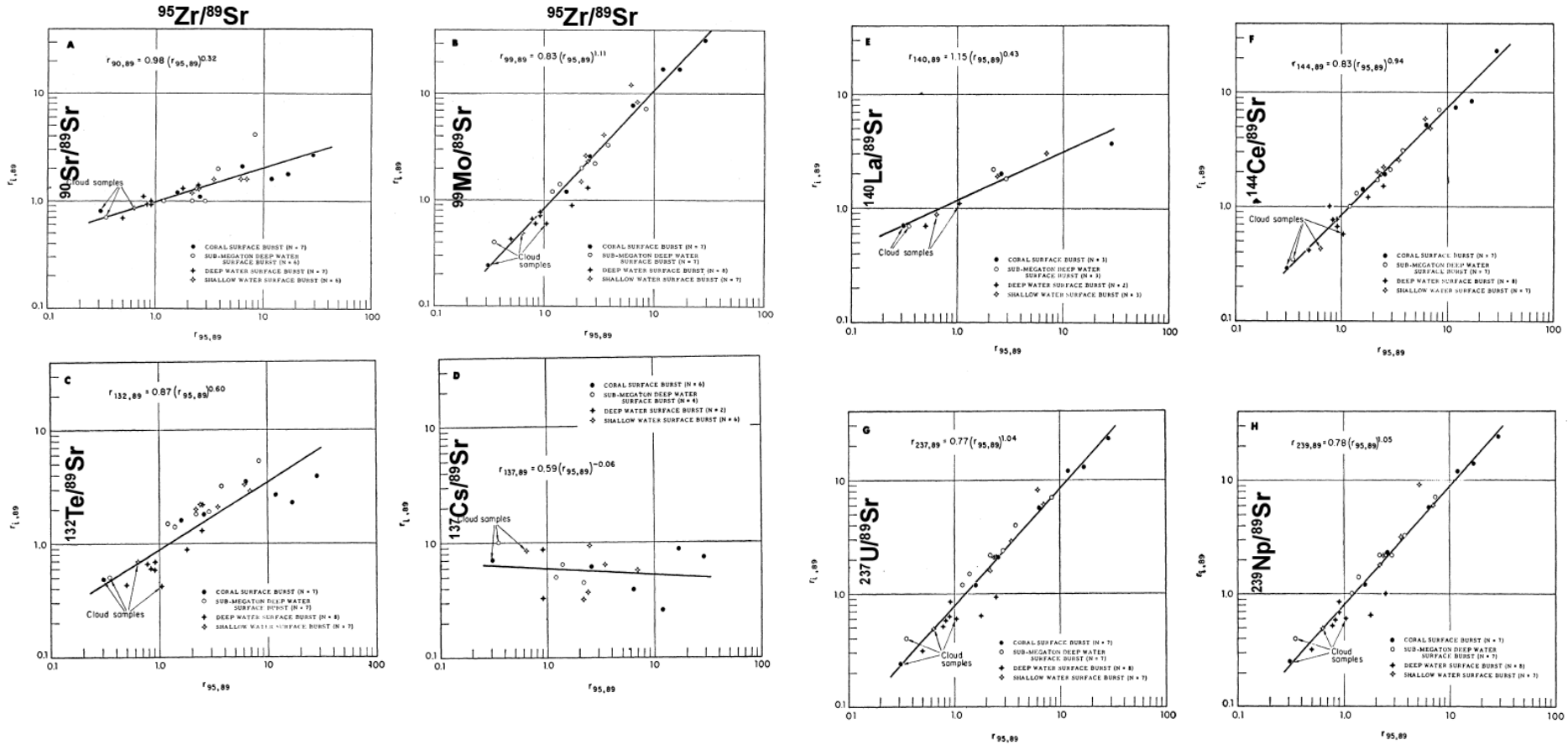
Laser Raman spectroscopy



More Diffusion Modeling



Freiling 1961



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